

Green County: Groundwater Quality

Kevin Masarik
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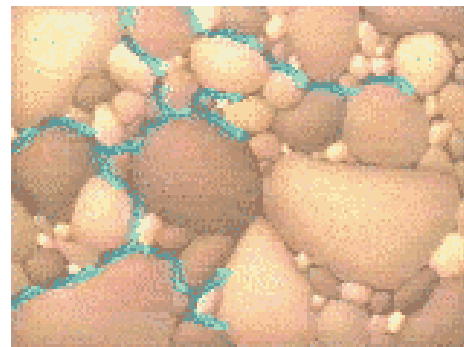
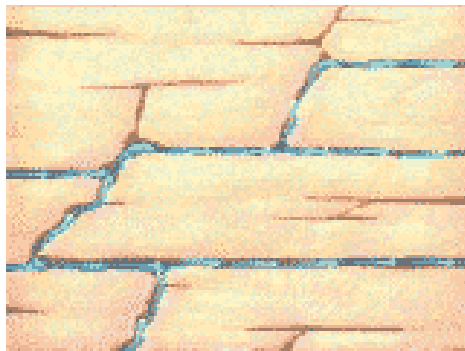
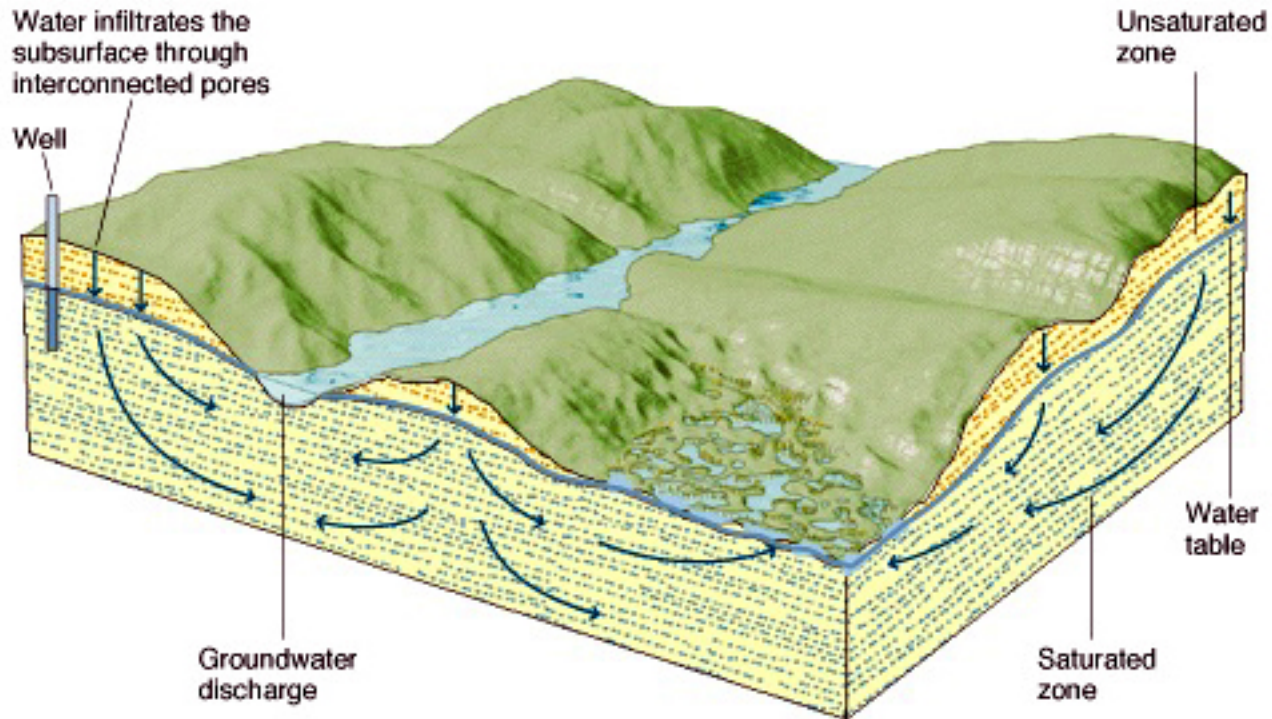


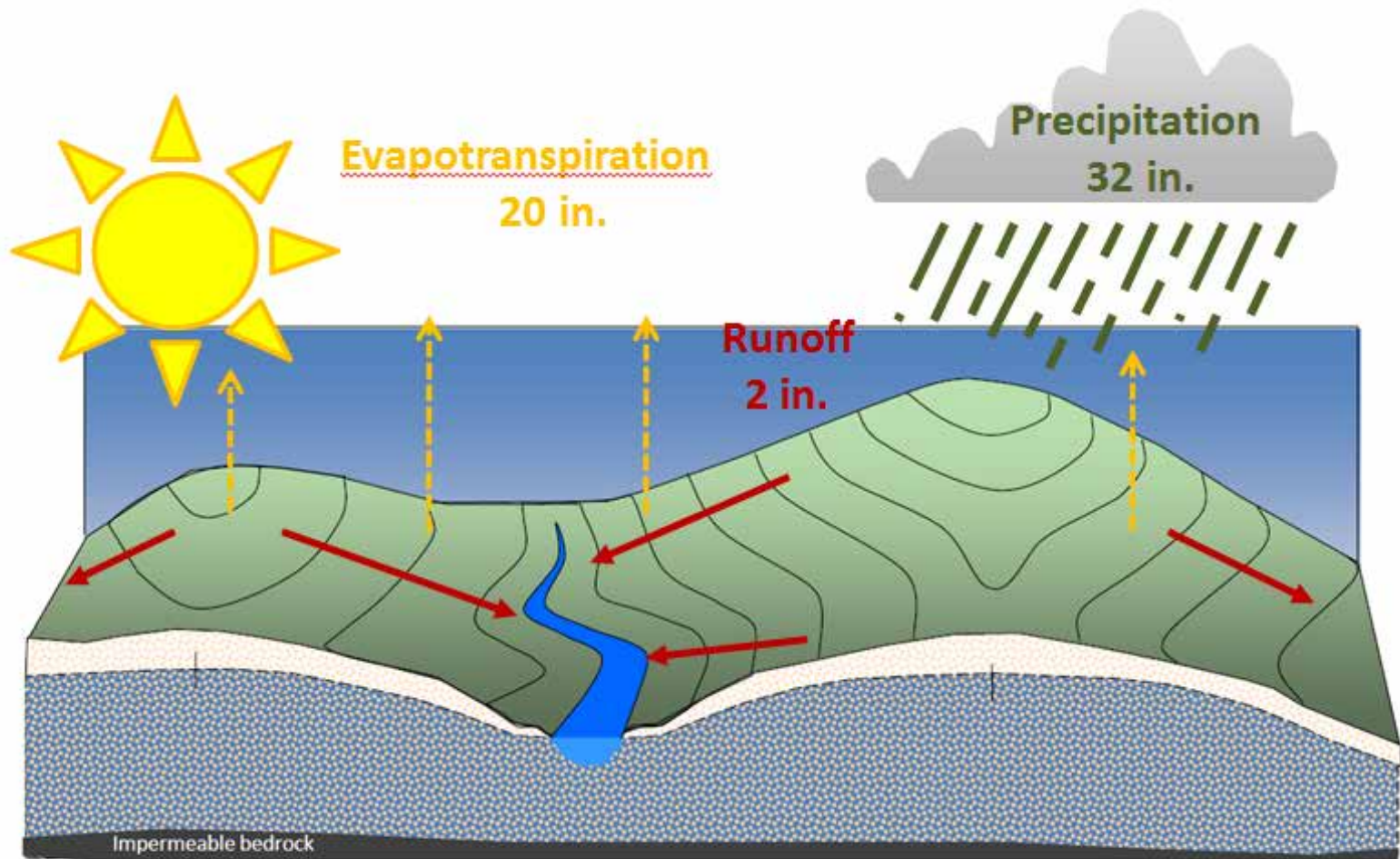
University of Wisconsin-Stevens Point
College of Natural Resources

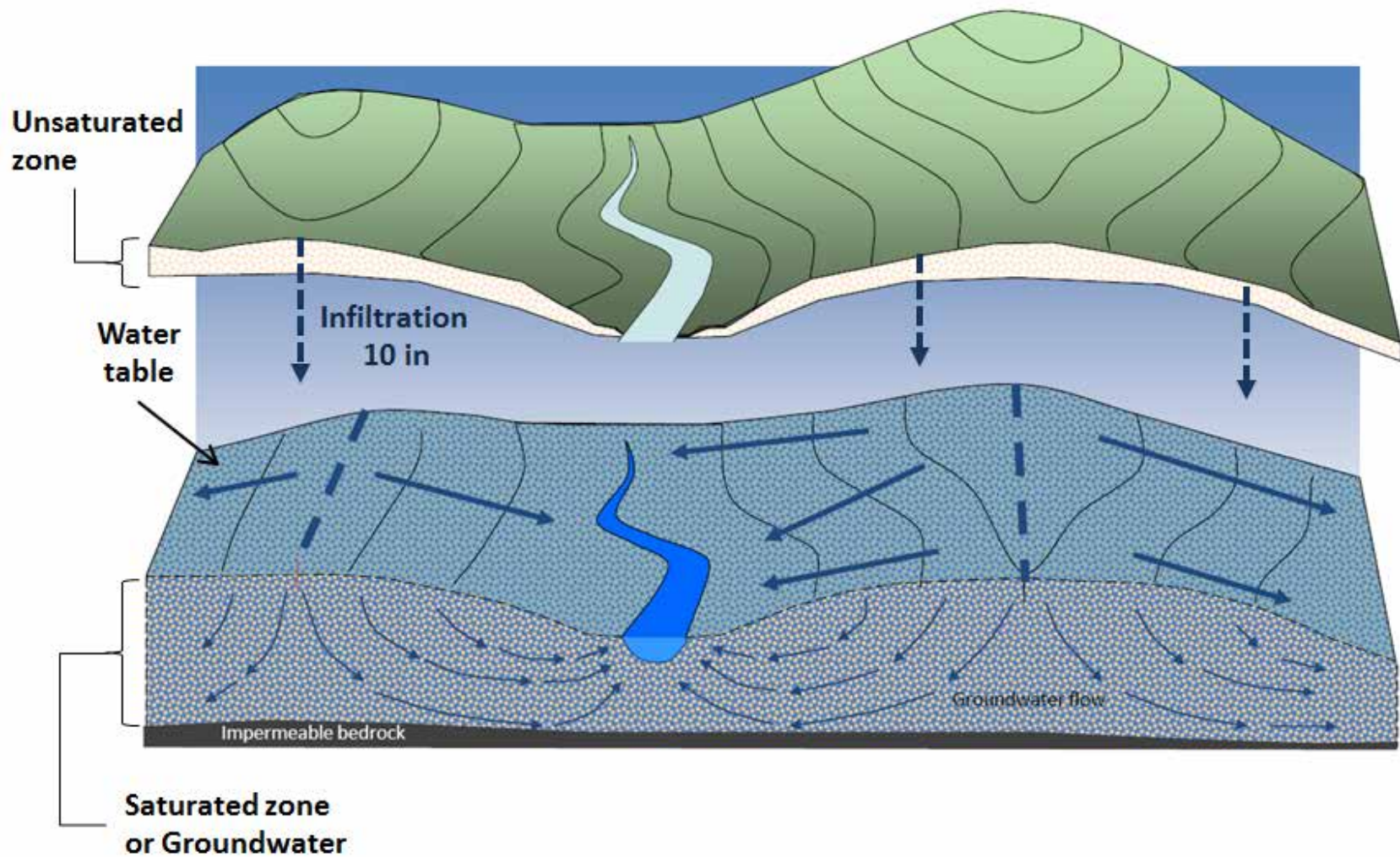


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Groundwater Movement





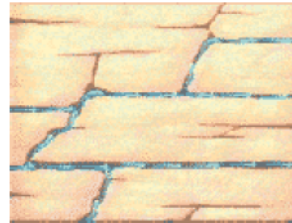


Aquifers: Our groundwater storage units

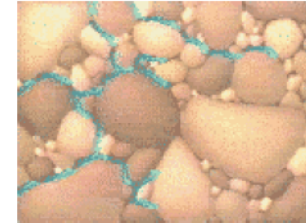
Aquifers are geologic formations that store and transmit groundwater.

The aquifer properties determine how quickly groundwater flows, how much water an aquifer can hold and how easily groundwater can become contaminated. Some aquifers may also contain naturally occurring elements that make water unsafe.

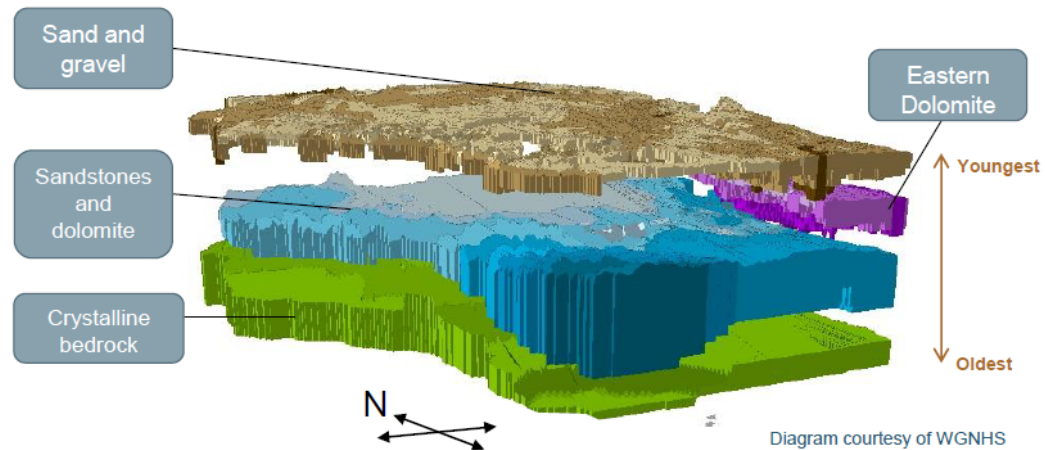
Wisconsin's geology is like a layered cake. Underneath all of Wisconsin lies the Crystalline bedrock which does not hold much water. Think of this layer like the foundation of your house. All groundwater sits on top of this foundation. Groundwater is stored in the various **sandstone**, **dolomite** and **sand/gravel** aquifers above the **crystalline bedrock** layer. The layers are arranged in the order which they formed, oldest on the bottom and youngest on top.



Water and contaminants can move quickly through cracks and fractures.



Water moving through tiny spaces in between sand particles or sandstone moves slower and allows for filtration of some contaminants.



BEDROCK GEOLOGY OF WISCONSIN

UNIVERSITY OF WISCONSIN-EXTENSION
Geological and Natural History Survey

APRIL 1981
REVISED 2005

EXPLANATION

DEVONIAN

D dolomite and shale

SILURIAN

Sd dolomite

ORDOVICIAN

Os Maquoketa Formation—shale and dolomite

St Siniperce Group—dolomite with some limestone and shale

Sp St. Peter Formation—sandstone with some limestone shale and conglomerate

Op Prairie du Chien Group—dolomite with some sandstone and shale

CAMBRIAN

C sandstone with some dolomite and shale

MIDDLE PROTEROZOIC

K Keweenaw rock—

ss, sandstone

v, basaltic to rhyolitic lava flows

t, gabbroic, anorthositic and granitic rock

Wolf River rock—

g, rapakivi granite, granite, and syenite

a, anorthositic and gabbro

LOWER PROTEROZOIC

q quartzite

g granite, diorite, and gneiss

ms, metasedimentary rock, argillite, siltstone, quartzite, greywacke, and iron formation

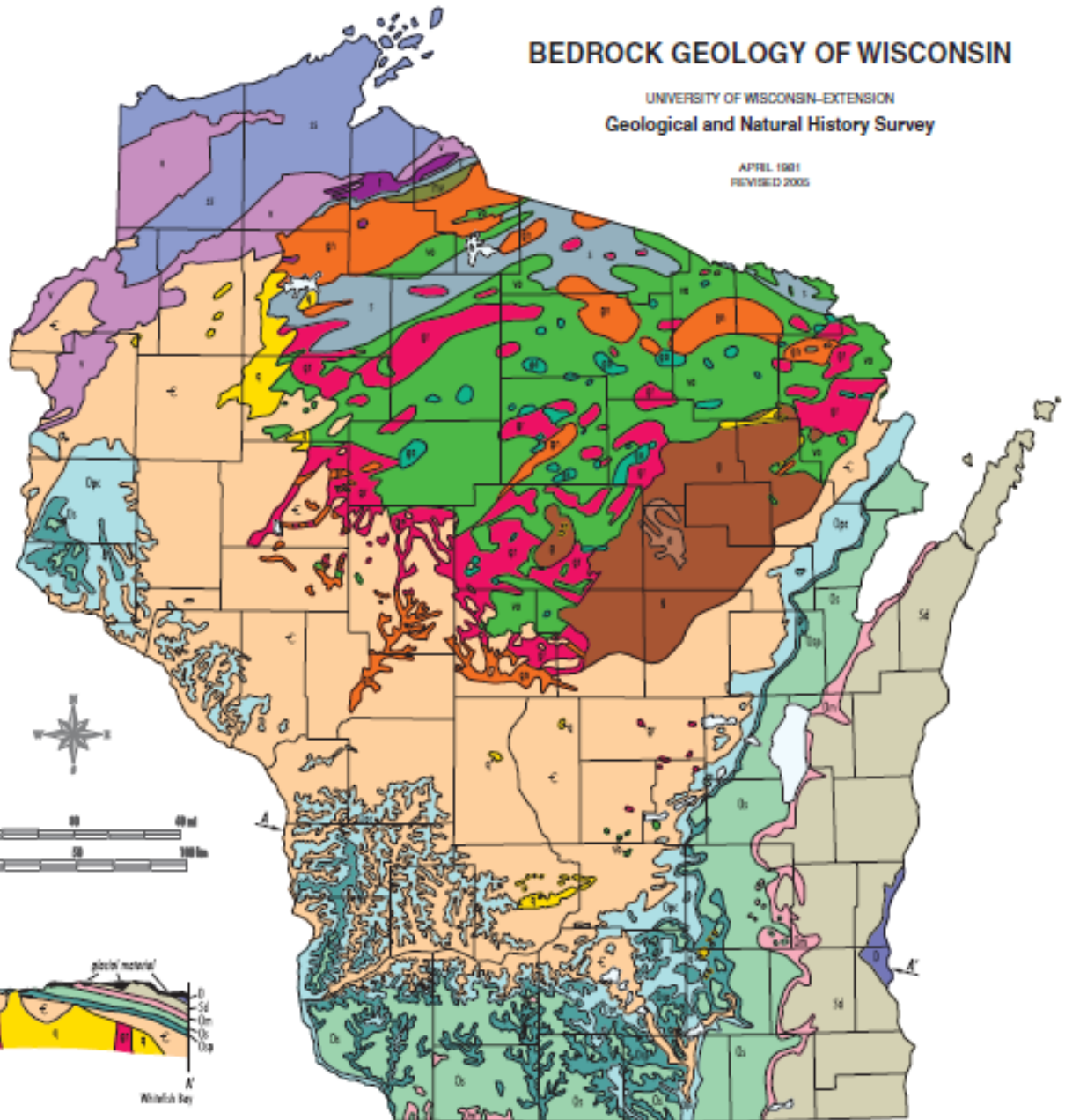
va, basaltic to rhyolitic metavolcanic rock with some metasedimentary rock

ga, meta-gabbro and hornblende diorite

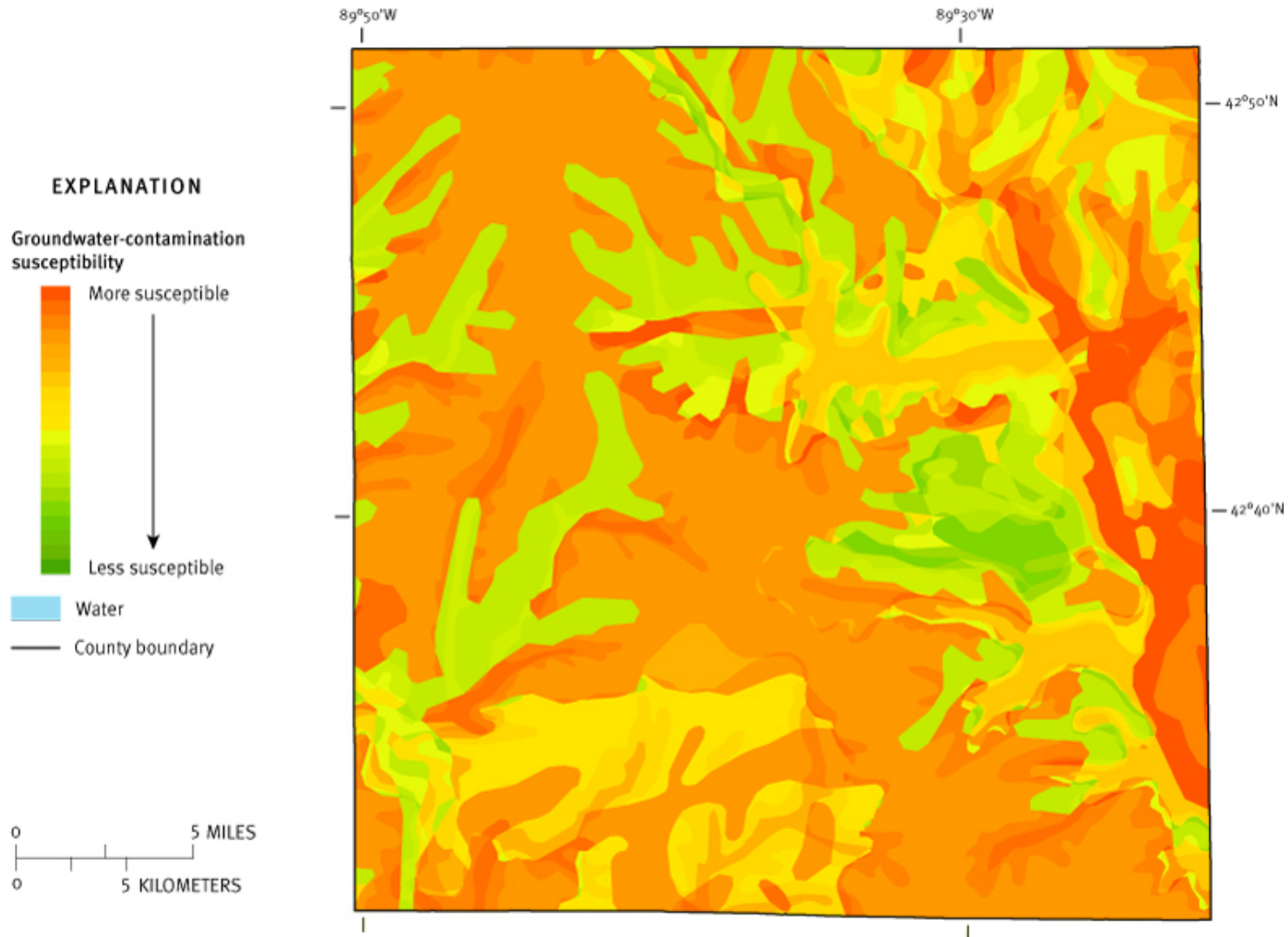
LOWER PROTEROZOIC OR UPPER ARCHEAN

mv, metavolcanic rock

gn, granite, gneiss, and amphibolite



Green County – Groundwater-Contamination Susceptibility Analysis

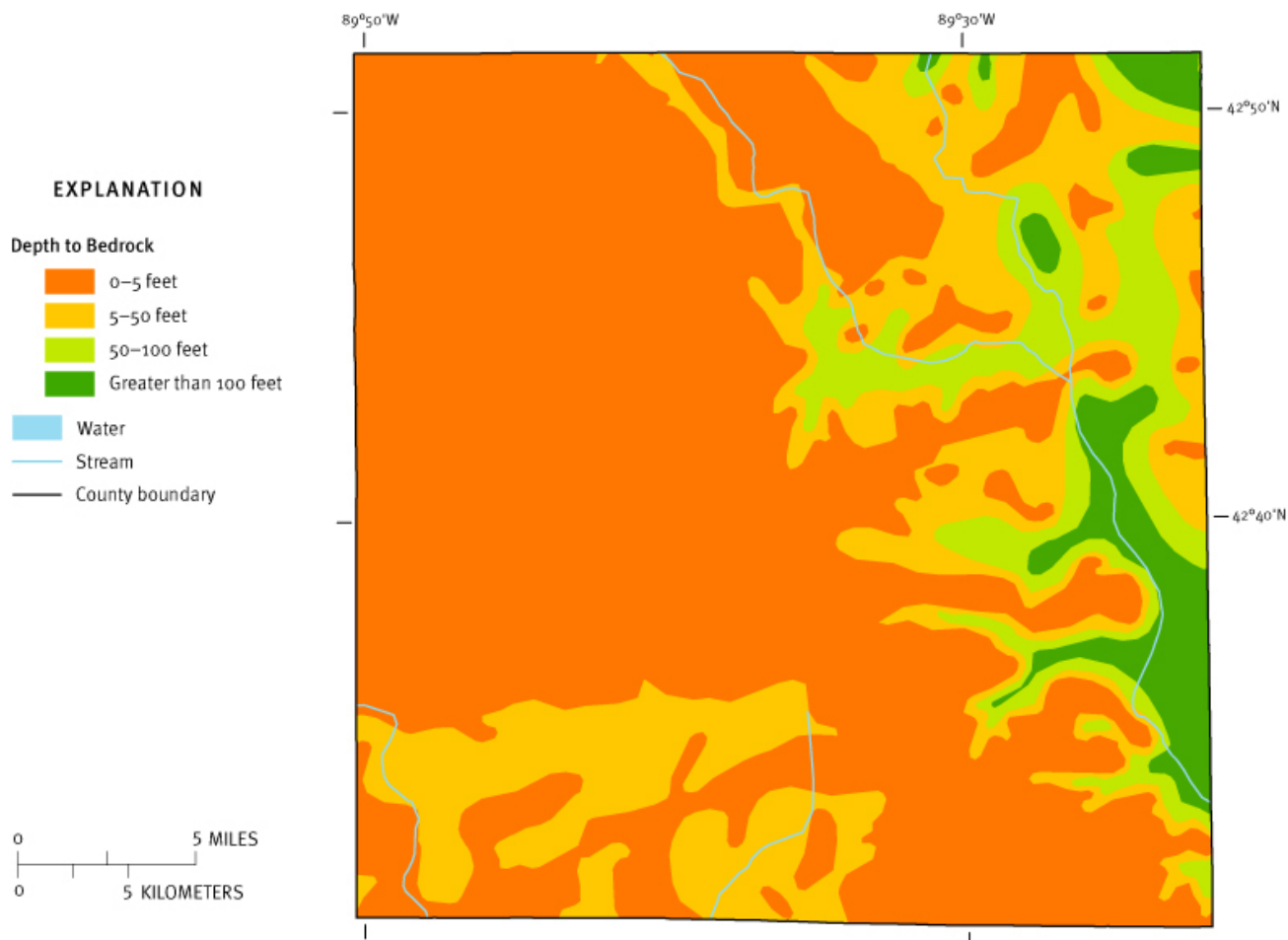


This groundwater-contamination susceptibility map is a composite of five resource characteristic maps, each of which was derived from generalized statewide information at small scales, and cannot be used for any site-specific purposes.

Map source: Schmidt, R.R., 1987, Groundwater contamination susceptibility map and evaluation: Wisconsin Department of Natural Resources, Wisconsin's Groundwater Management Plan Report 5, PUBL-WR-177-87, 27 p.

Figure created for the "Protecting Wisconsin's Groundwater Through Comprehensive Planning" web site, 2007, <http://wi.water.usgs.gov/gwcomp/>

Green County – Depth to Bedrock

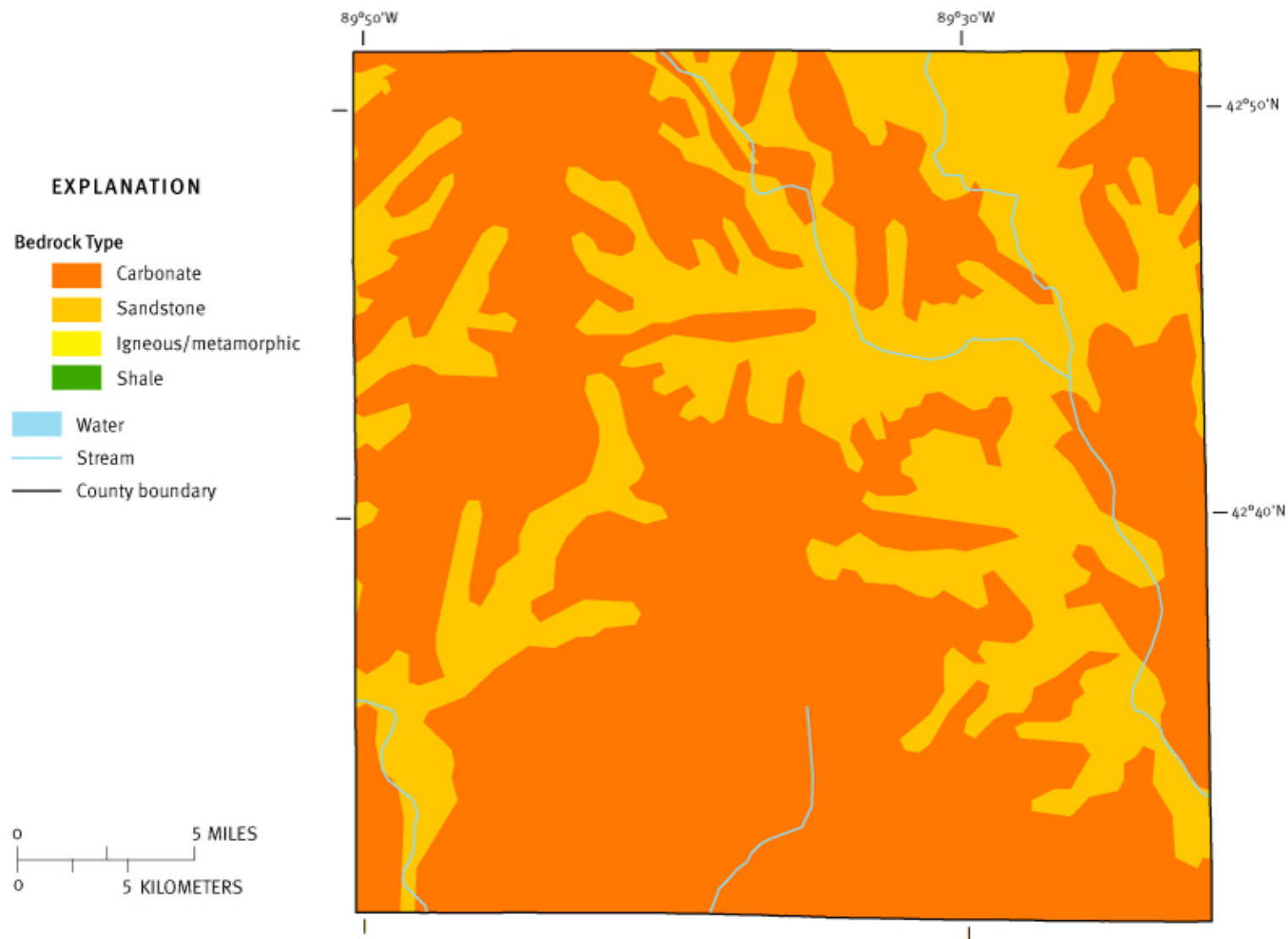


This resource characteristic map was derived from generalized statewide information at small scales, and cannot be used for any site-specific purposes.

Map source: Schmidt, R.R., 1987, Groundwater contamination susceptibility map and evaluation: Wisconsin Department of Natural Resources, Wisconsin's Groundwater Management Plan Report 5, PUBL-WR-177-87, 27 p.

Figure created for the "Protecting Wisconsin's Groundwater Through Comprehensive Planning" web site, 2007, <http://wi.water.usgs.gov/gwcomp/>

Green County – Bedrock Type



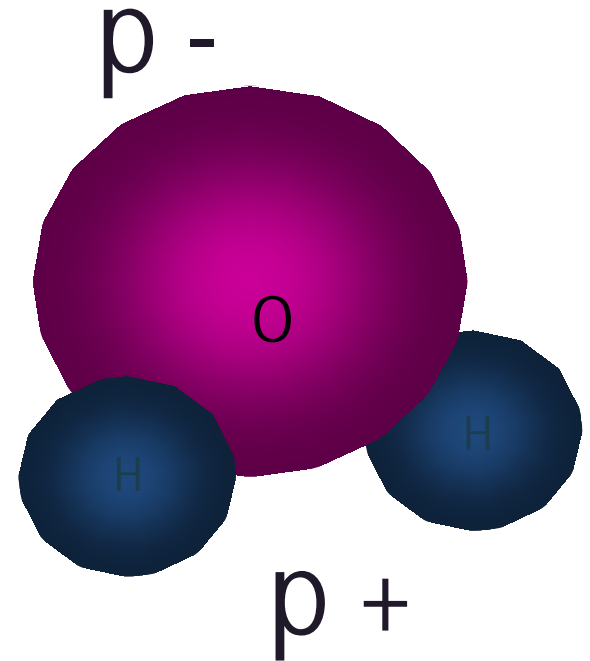
This resource characteristic map was derived from generalized statewide information at small scales, and cannot be used for any site-specific purposes.

Map source: Schmidt, R.R., 1987, Groundwater contamination susceptibility map and evaluation: Wisconsin Department of Natural Resources, Wisconsin's Groundwater Management Plan Report 5, PUBL-WR-177-87, 27 p.

Figure created for the "Protecting Wisconsin's Groundwater Through Comprehensive Planning" web site, 2007, <http://wi.water.usgs.gov/gwcomp/>

water quality basics

- “Universal Solvent”
- Naturally has “stuff” dissolved in it.
 - Impurities depend on rocks, minerals, land-use, plumbing, packaging, and other materials that water comes in contact with.
- Can also treat water to take “stuff” out



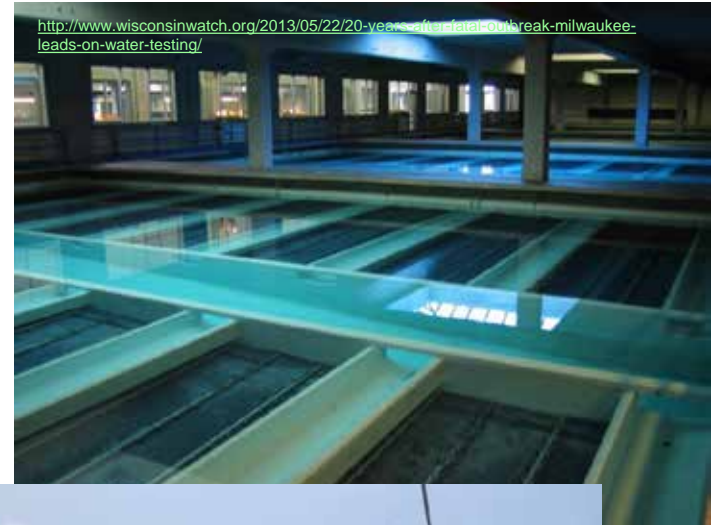
Public vs. Private Water Supplies

Public Water Supplies

- Regularly tested and regulated by drinking water standards.

Private Wells

- Not required to be regularly tested.
- Not required to take corrective action
- Owners must take special precautions to ensure safe drinking water.



Coliform bacteria

§ Generally do not cause illness, but indicate a pathway for potentially harmful microorganisms to enter your water supply.

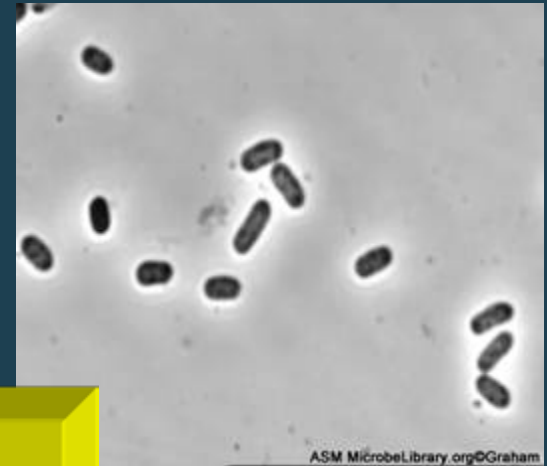
§ Harmful bacteria and viruses can cause gastrointestinal disease, cholera, hepatitis

§ Sanitary water supply should not contain any coliform bacteria

§ Recommend using an alternative source of water until a test indicates your well is absent of coliform bacteria

§ Sources:

- Live in soils and on vegetation
- Human and animal waste
- Sampling error



ASM MicrobeLibrary.org©Graham

Present = Unsafe

Absent = Safe

If coliform bacteria was detected, the sample is checked for E.coli

- Ø Confirmation that bacteria originated from a human or animal fecal source.
- Ø E. coli are often present with harmful bacteria, viruses and parasites that can cause serious gastrointestinal illnesses.
- Ø Any detectable level of E.coli means your water is unsafe to drink.

Information Sources: United States Department of Health and Human Services – Centers for Disease Control and Prevention (www.cdc.gov) and United States Environmental Protection Agency (www.epa.gov)

Contaminants	Sources	Symptoms
BACTERIA		
<i>Escherichia coliform (E. coli)</i> <i>Salmonella</i> <i>Campylobacter</i> <i>E. coli</i> 0157 (Requires a special water test for detection. Causes similar, but more serious illness than other E.coli strains. Requires medical treatment.)	<ul style="list-style-type: none"> • Infected human and animal feces • Manure • Septic systems • Sewage 	<ul style="list-style-type: none"> • Gastrointestinal illness • Low-grade fever • Begins 12 hrs - 7 days after exposure
<i>Leptosporidia</i>	<ul style="list-style-type: none"> • Urine of livestock, dogs and wildlife • Manure 	<ul style="list-style-type: none"> • High fever, severe headache and red eyes • Gastrointestinal illness • Begins 2-28 days after exposure
MICROSCOPIC PARASITES		
<i>Cryptosporidia</i> <i>Giardia</i>	<ul style="list-style-type: none"> • Infected human and animal feces • Manure • Septic systems • Sewage 	<ul style="list-style-type: none"> • Gastrointestinal illness • Begins 2-14 days after exposure
VIRUSES		
Norovirus	<ul style="list-style-type: none"> • Infected human feces and vomit • Septic systems • Sewage 	<ul style="list-style-type: none"> • Gastrointestinal illness • Low-grade fever & headache • Begins 12-48 hrs after exposure
CHEMICALS		
Nitrate	<ul style="list-style-type: none"> • Fertilizers • Manure • Bio-solids • Septic systems 	Methemoglobinemia or "Blue Baby Syndrome" – No documented cases in Door County, but elevated nitrate levels in well water may indicate risk of contamination by additional pathogens.
Atrazine (trade-name herbicide for control of broadleaf and grassy weeds)	Estimated to be most heavily used herbicide in the U.S. in 1987/89, with its most extensive use for corn and soybeans in the Midwest, including WI. In 1993, it became a restricted-use herbicide nationally. U.S. EPA set a max. contaminant level (MCL) at 3 parts per billion for safe drinking water.	Short-term exposure above the MCL may cause: congestion of heart, lungs and kidneys; low blood pressure; muscle spasms; weight loss; damage to adrenal glands. Long-term exposure above MCL may cause: weight loss, cardiovascular damage, retinal and some muscle degeneration; cancer.

Well Construction

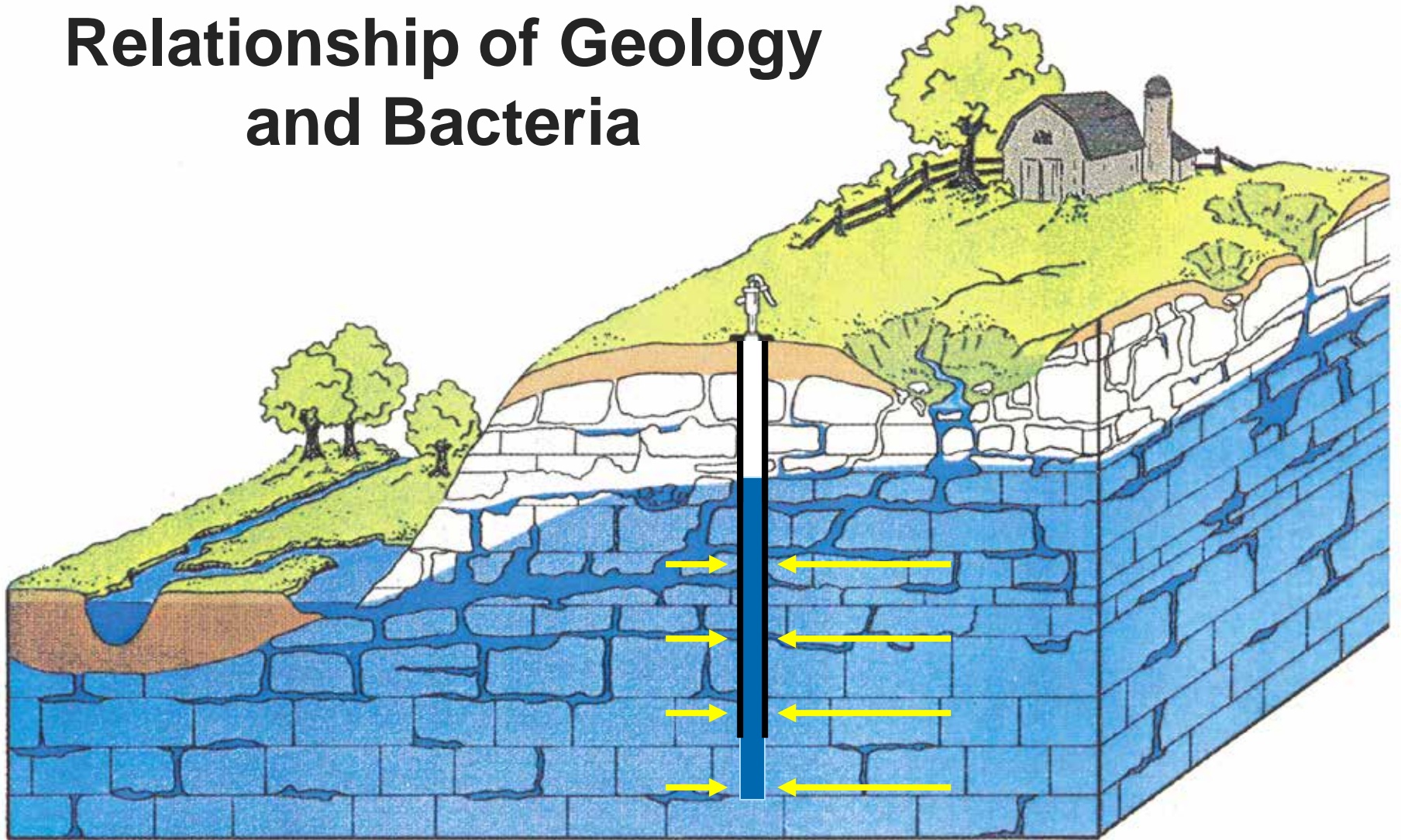


Photo: Sandy Heimke, WI DNR

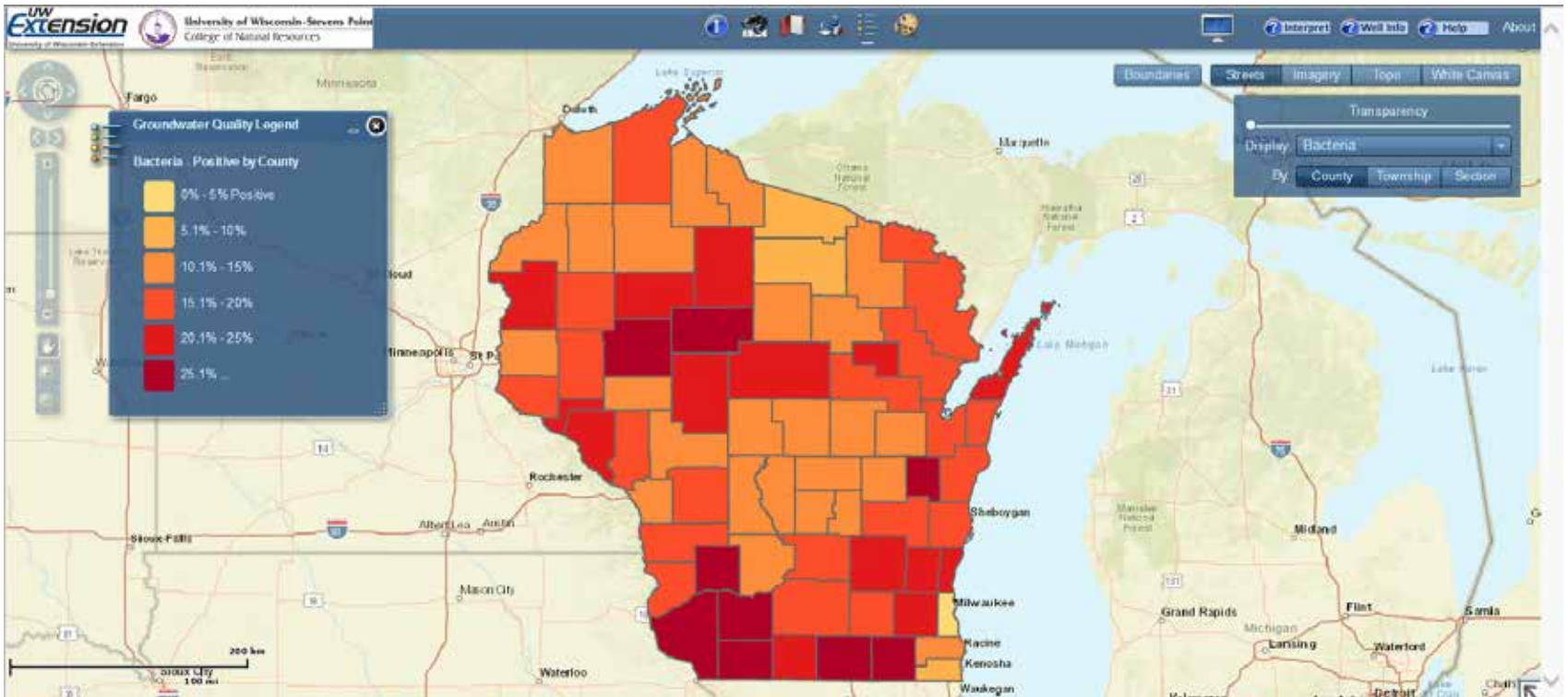


Photos courtesy of: Matt Zoschke

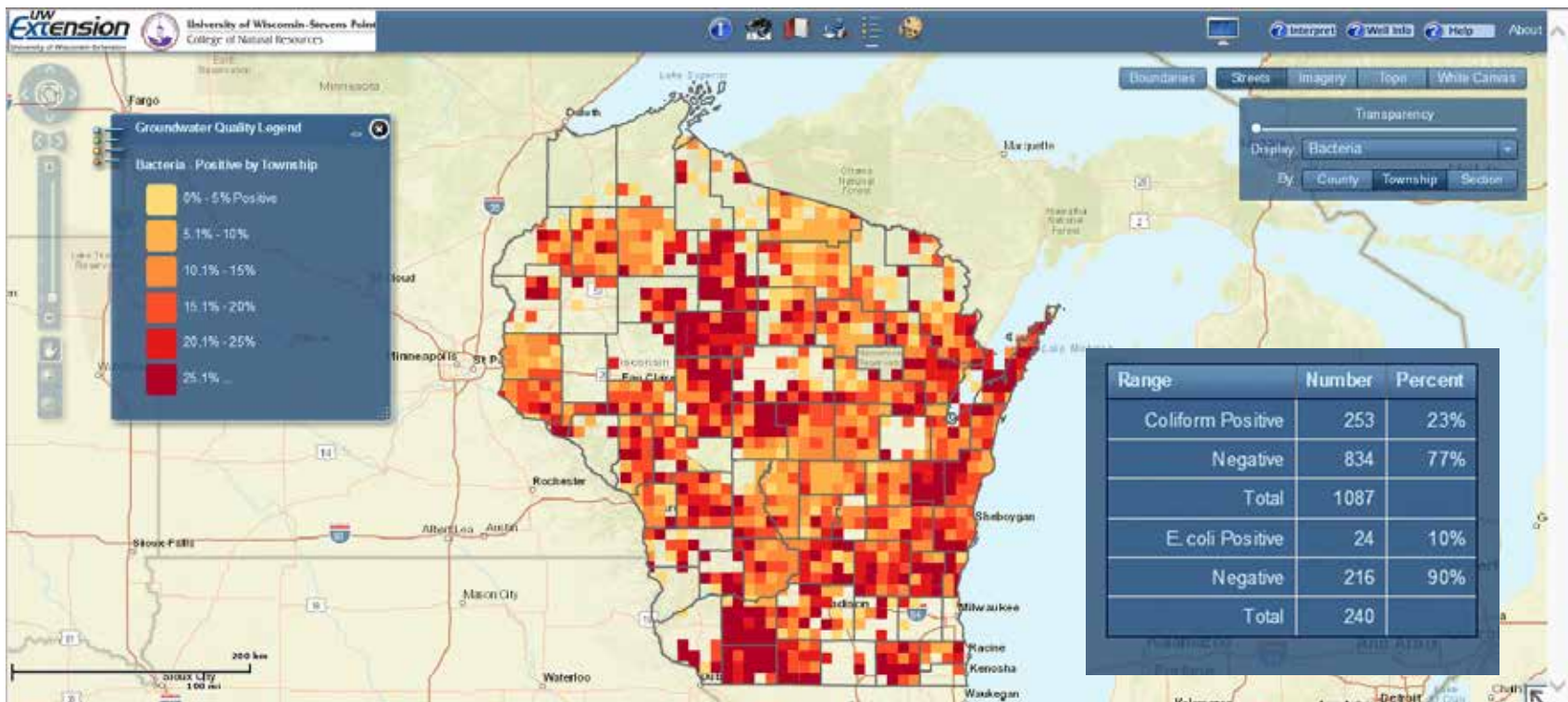
Relationship of Geology and Bacteria



Coliform Bacteria by County



Coliform Bacteria by County



Nitrate and Human Health

Infants and pregnant women

- Methemoglobinemia or “blue-baby syndrome”
- Possible correlation to central nervous system malformations

Adults

Possible correlations to:

- Non-Hodgkin’s lymphoma
- Various cancers (ex. gastric, bladder)
- Thyroid function
- Diabetes in children

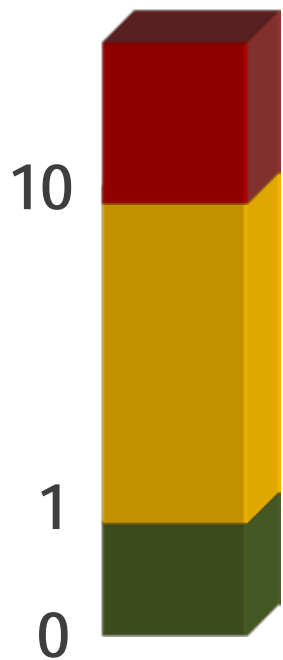


- *Many are statistical studies that provide correlation between nitrate and health problems
- *Studies don’t always agree, but cannot say with certainty that nitrate poses no health risk.

Nitrate often indicator of other possible contaminants
(ex. other agricultural contaminants, septic effluent, etc.)

[Wisconsin Groundwater Coordinating Council, 2015](#); [Weyer, 1999](#)

Nitrate in drinking water



- **Greater than 10 mg/L**
Impacted at a level that exceeds state and federal limits for drinking water

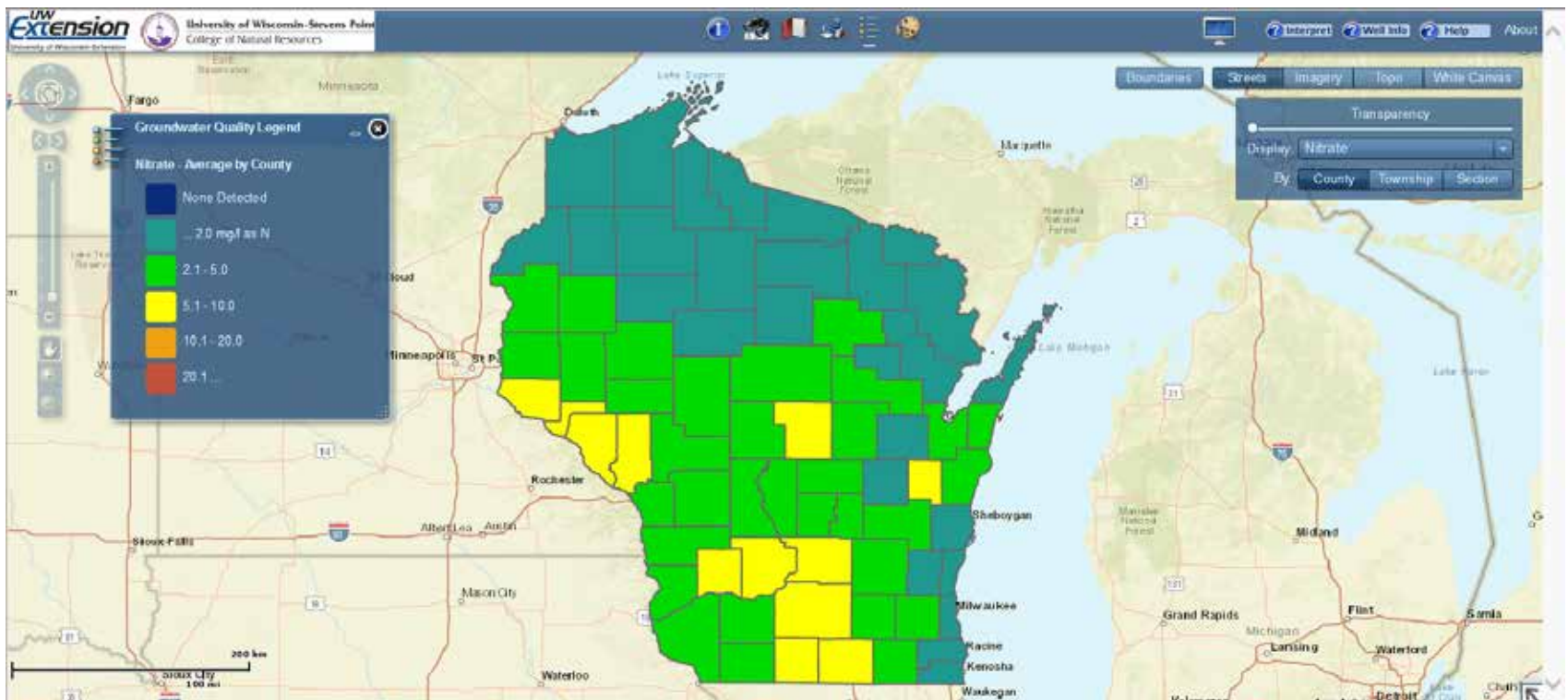
- DO NOT give water to infants
- DO NOT consume if you are a woman who is pregnant or trying to conceive
- RECOMMEND everyone avoid long-term consumption

- **Between 1 and 10 mg/L**
Evidence of land-use impacts

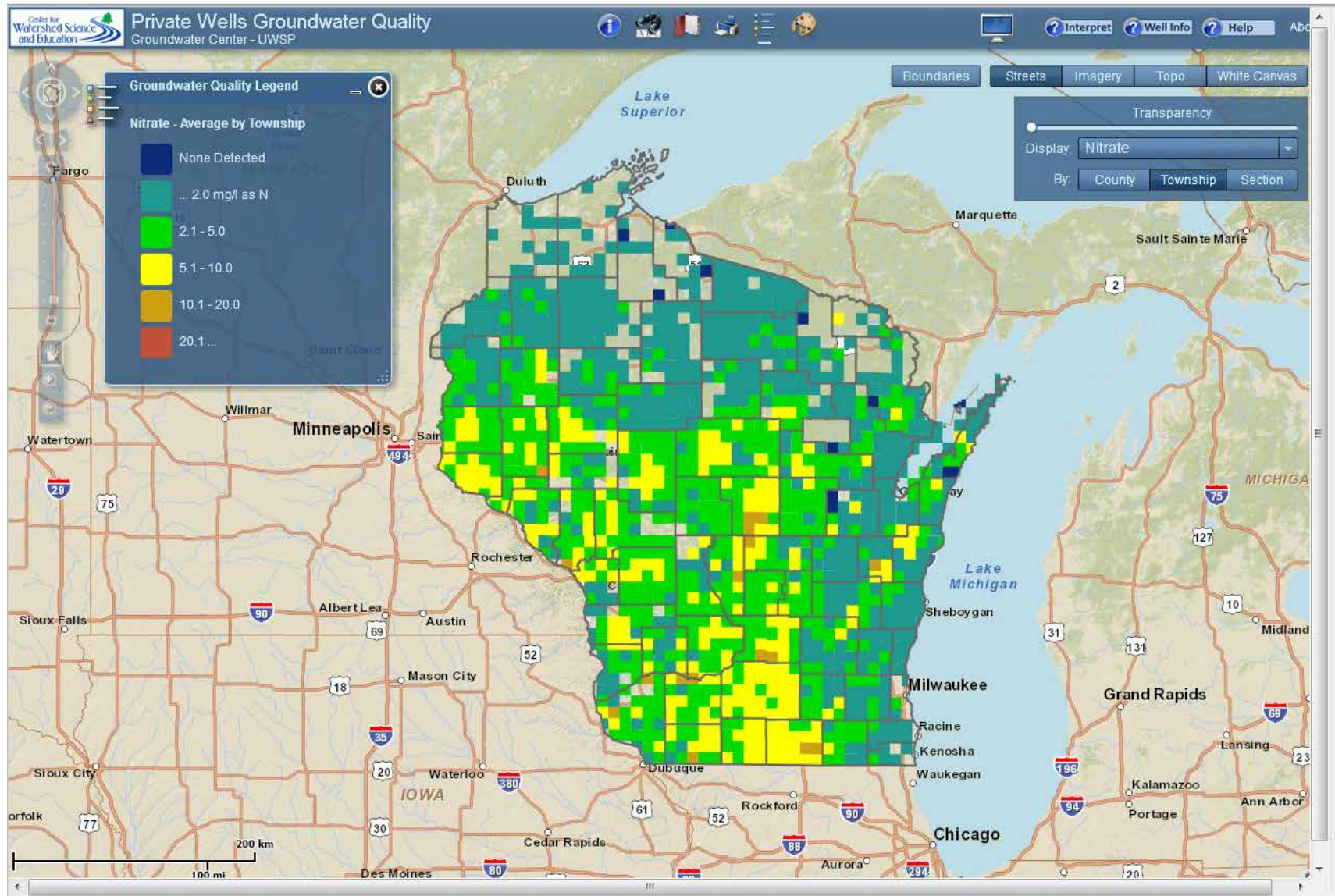
Considered suitable for drinking water

- **Less than 1 mg/L**
Natural or background levels in WI groundwater

Nitrate-Nitrogen Average by County

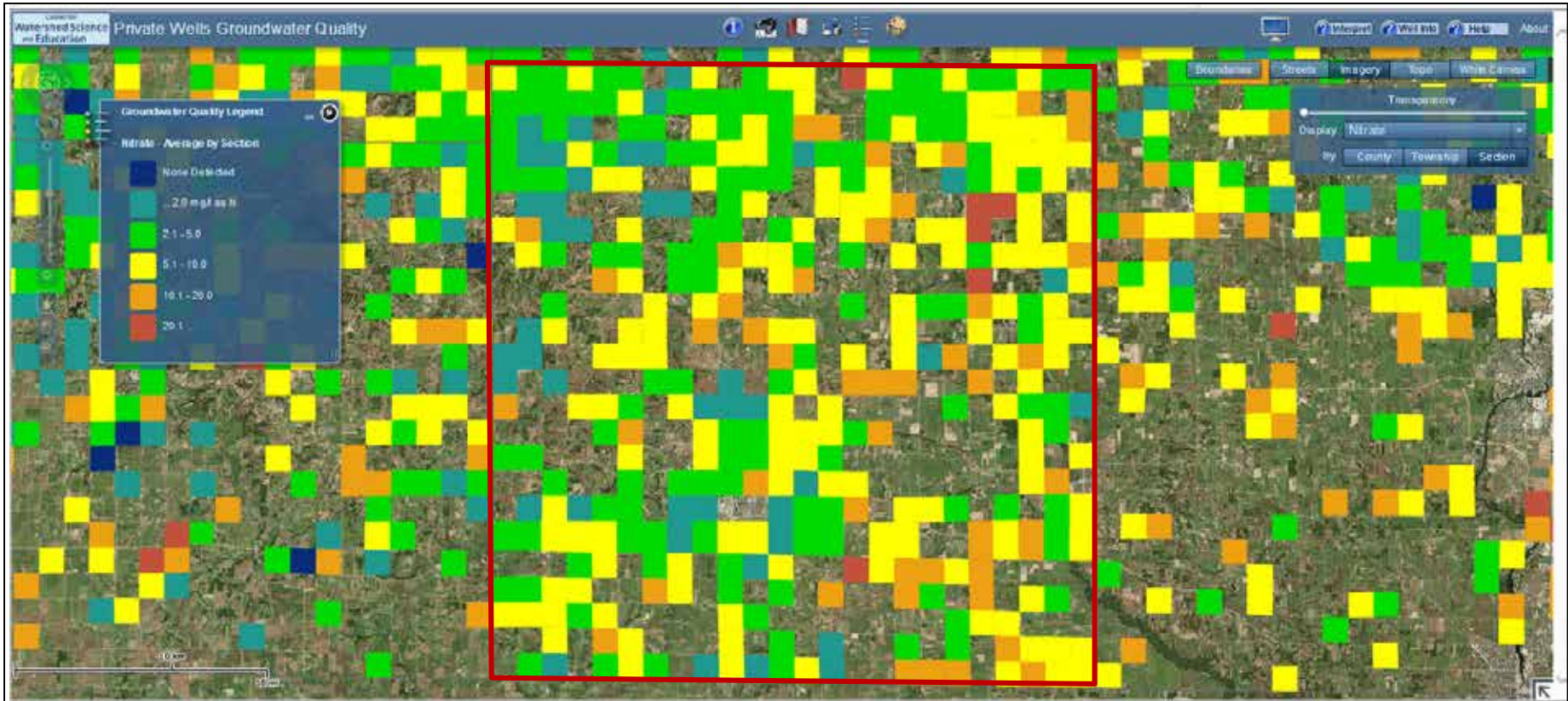


Nitrate-Nitrogen by Township-Range



<http://www.uwsp.edu/cnr-ap/watershed/Pages/wellwaterviewer.aspx>

Average Nitrate-N concentration by section.



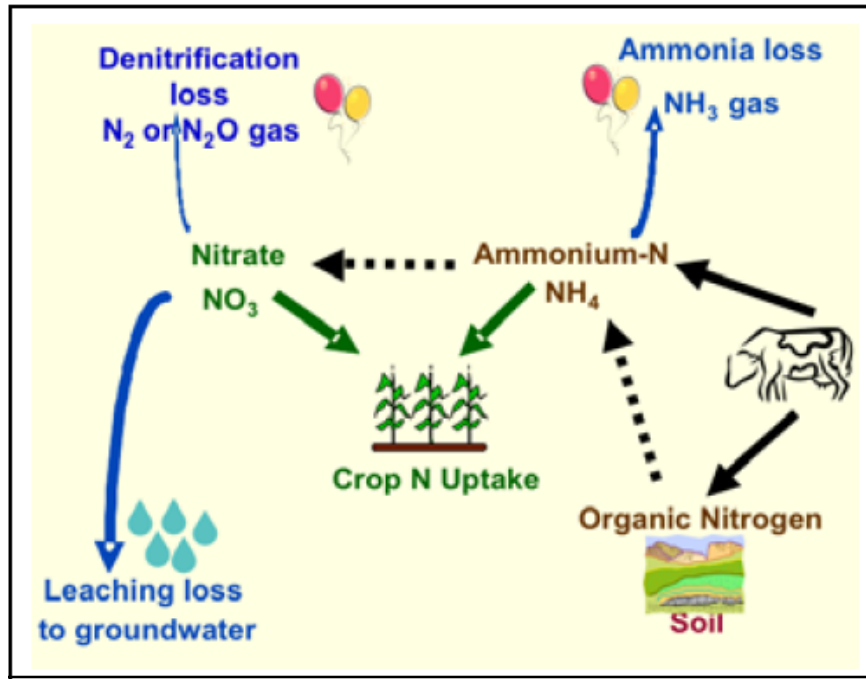
Range	Number	Percent	Summary
None Detected	194	7%	Minimum: No Detect
... 2.0	622	23%	
2.1 - 5.0	724	27%	Median: 4.1
5.1 - 10.0	694	26%	Average: 5.61288
10.1 - 20.0	364	14%	
20.1 ...	64	2%	Maximum: 69.9
Total	2662		
> 10mg/l N	428	16%	Exceeds Health Standard

<http://www.uwsp.edu/cnr-ap/watershed/Pages/wellwaterviewer.aspx>

Green County
Nitrate Summary

Nitrogen Cycle

"Nitrogen is neither created nor destroyed"



<http://ir.library.oregonstate.edu/xmlui/bitstream/handle/1957/20528/em8954-e.pdf>

The Environment and N Loss from Manures—Why Do We Care?

Plant-available N (PAN) losses from the soil represent lost fertilizer value. Nitrogen can be lost as ammonia, nitrate, or nitrous oxides (Figure 1, page 3). Besides losing a valuable resource, the lost PAN can contribute to off-site problems.

Ammonia lost to the atmosphere is an air pollution problem in some areas of the western U.S., particularly in winter when atmospheric inversions prevent air mixing. In the atmosphere, ammonia can react with dust and other compounds to reduce visibility and to acidify rain or fog. Ammonia emissions may contribute to:

- Human health problems (inhalation hazard)
- Changes in natural plant communities in forests and rangeland. (Nitrogen deposited in N-poor ecosystems can alter the balance between adapted species and N-loving invasive species.)
- Acid fog or rain damage to limestone buildings or cultural artifacts (for example, petroglyphs on limestone)
- Reduction in visibility (haze)

Nitrate moves with soil water. Nitrate lost from soil enriches groundwater or surface water and can contribute to:

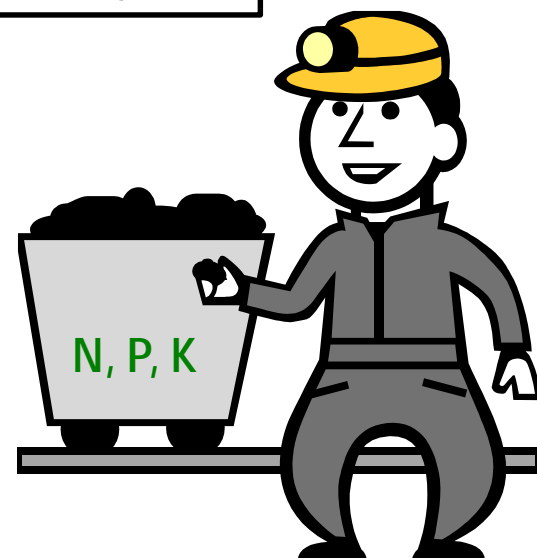
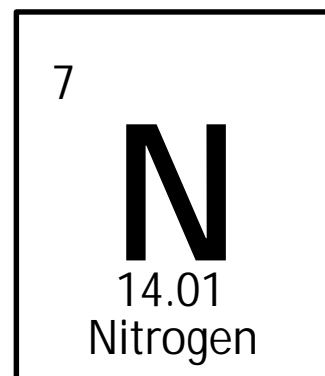
- Human health problems (blue baby syndrome, elevated cancer risk)
- Algae blooms in lakes or other slow-moving bodies of water
- Reduced survival and reproduction of some amphibians

Nitrous oxides lost to the atmosphere through denitrification can contribute to:

- Human health problems (inhalation hazard)
- Global warming (A molecule of nitrous oxide (N_2O) traps approximately 300 times more heat than a molecule of carbon dioxide.)
- Increased N deposits in sensitive ecosystems, resulting in soil acidification or change in plant communities
- Reduction in visibility (haze)

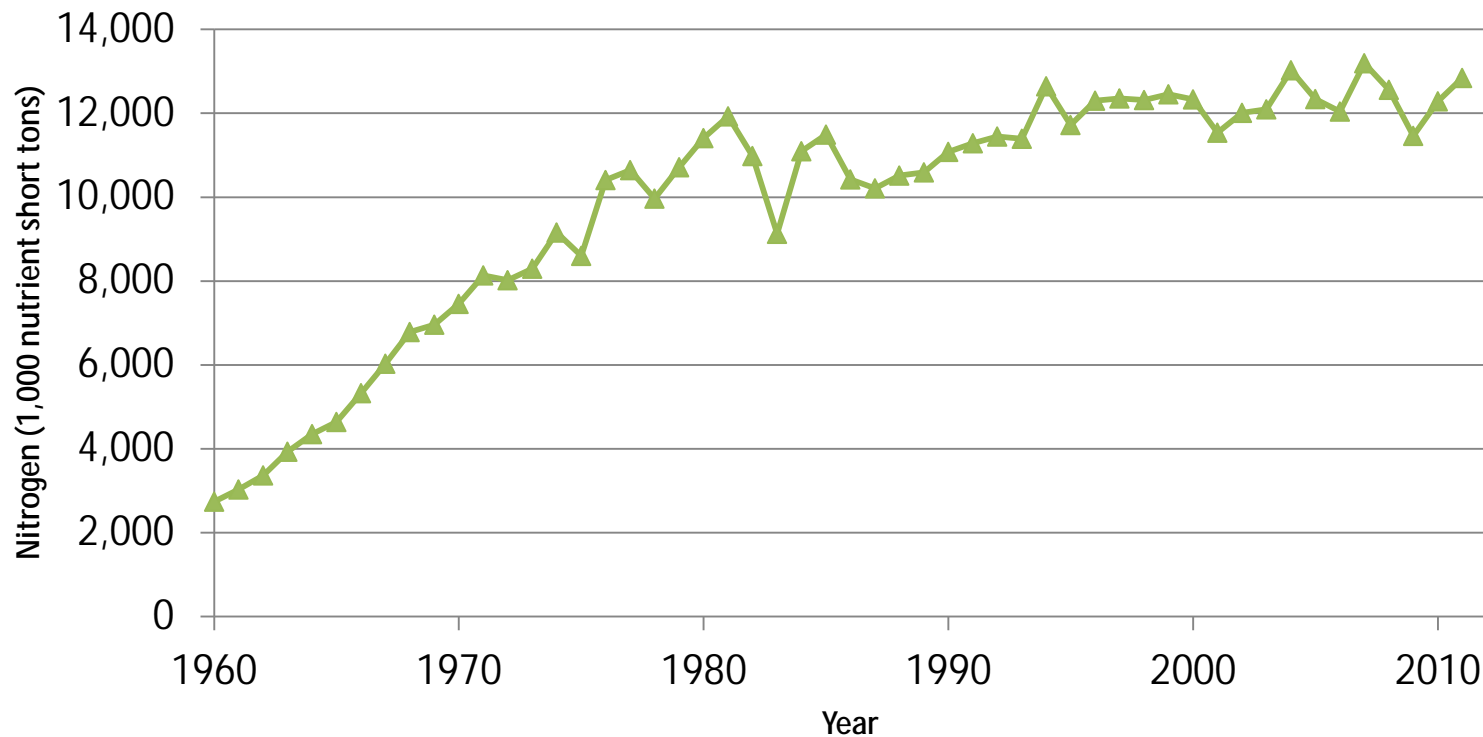
Nitrogen is vital to agriculture

- § Ancient civilizations farmed fertile flood plains
- § Animal manures
- § Crop rotations w/legumes
- § Prairies and other organic rich soils
- § Industrial fixation of N leads to commercial fertilizer and dramatic increase in N applications
- § Manure management challenging



Historical Nitrogen Use

U.S. Consumption of Nitrogen Fertilizer (1960-2011)



USDA Fertilizer Use and Price, 2013

<http://www.ers.usda.gov/data-products/fertilizer-use-and-price.aspx>

Less

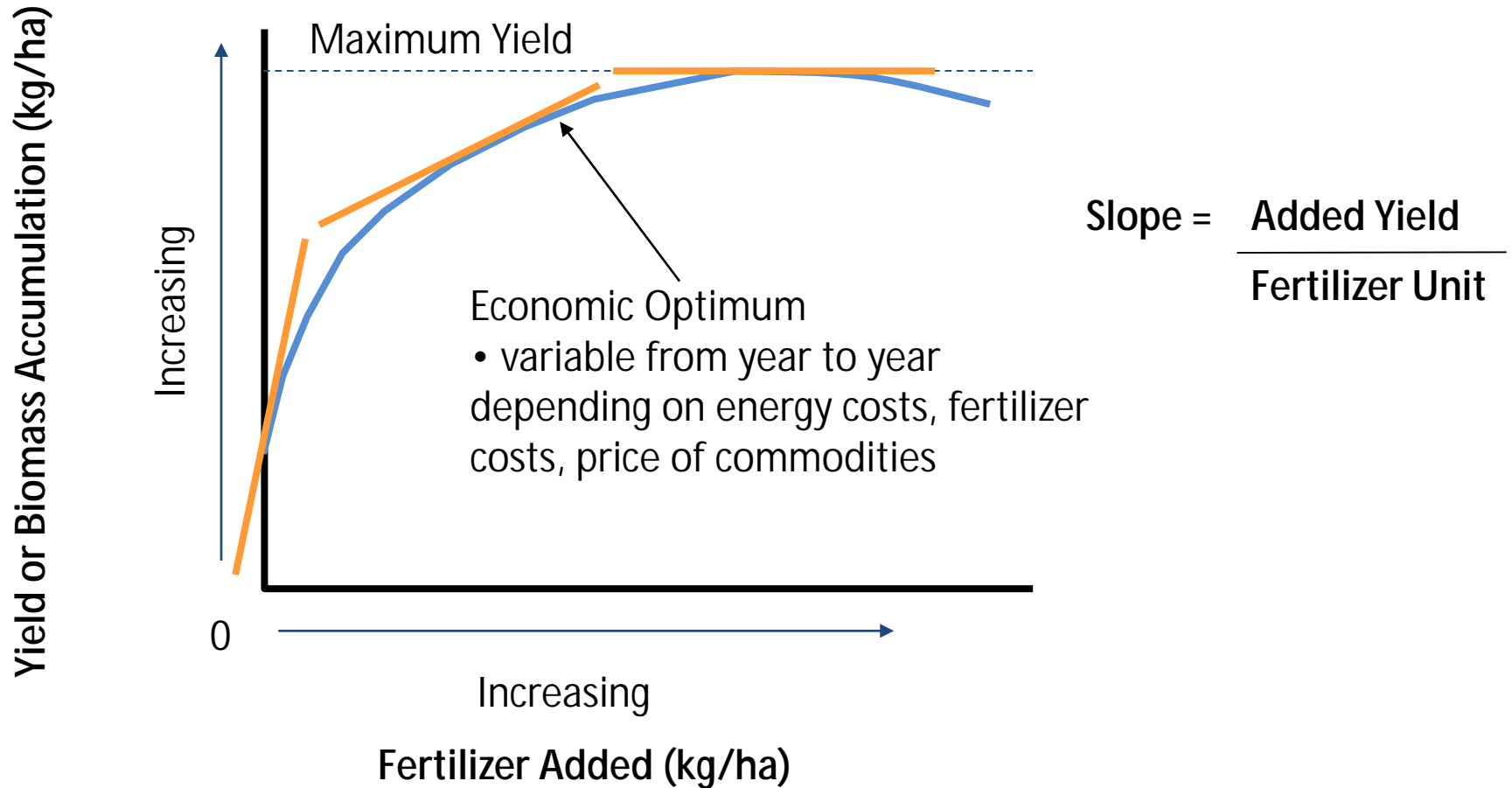
More



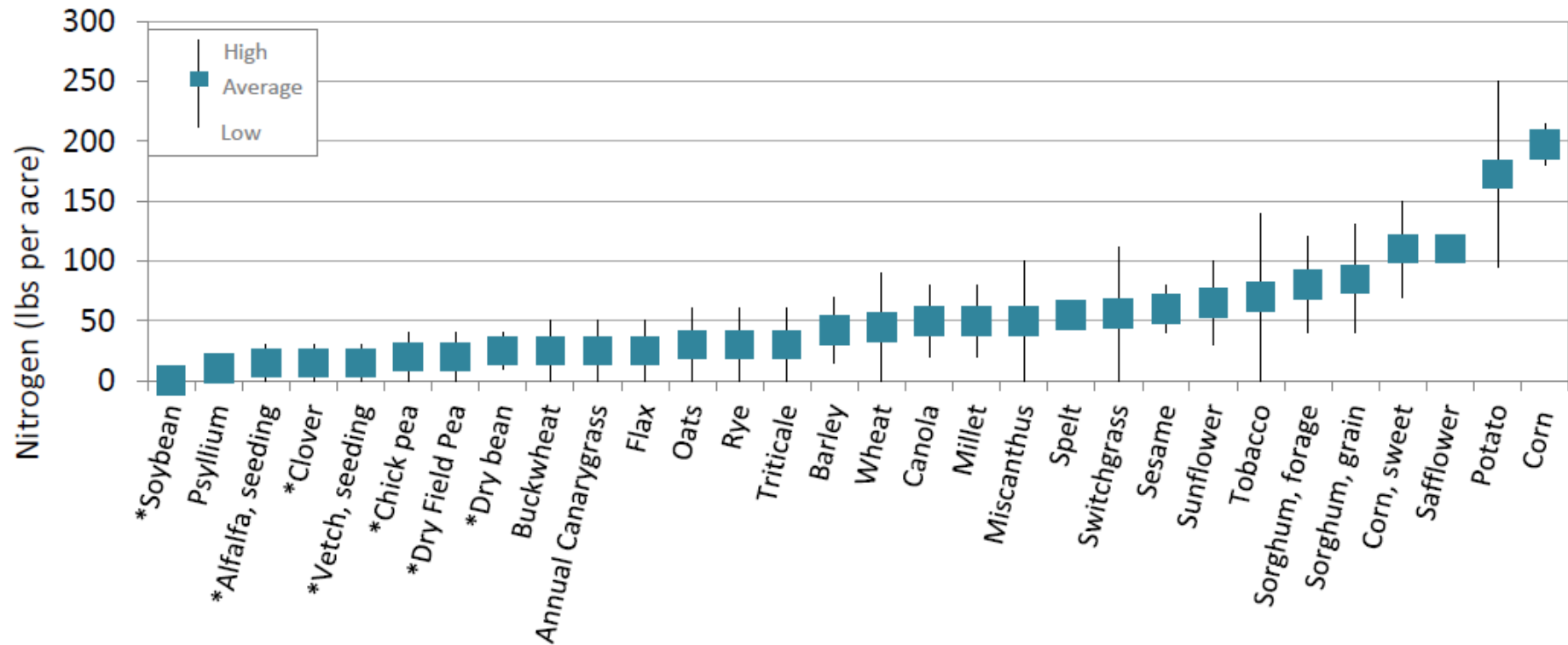
Nitrogen Fertilizer Added (lb/acre)



Yield response to nitrogen



Nitrogen fertilizer recommendations for common crops



* Legumes have symbiotic relationship with N fixing bacteria

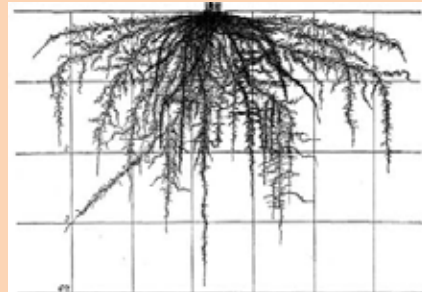
Alternative Field Crops Manual, 1989. University of Minnesota and University of Wisconsin -Madison
[Nutrient application guidelines for field, vegetable and fruit crops in Wisconsin. A2809](#). 2012. University of Wisconsin-Madison
 Miscanthus and switchgrass recommendations: Anderson et al., 2013; McIsaac et al., 2010; Vogel et al., 2002; Arundale et al, 2014

Comparing Corn to Perennial Ecosystems

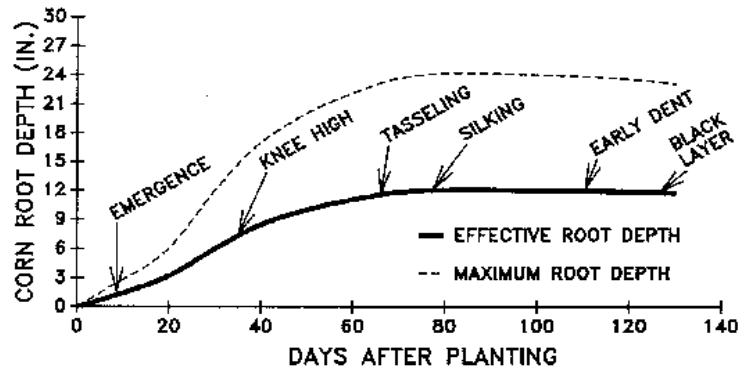
Nitrogen fertilizer use
efficiency for
Midwestern corn
systems

37%

([Cassman et. al. 2002](#))

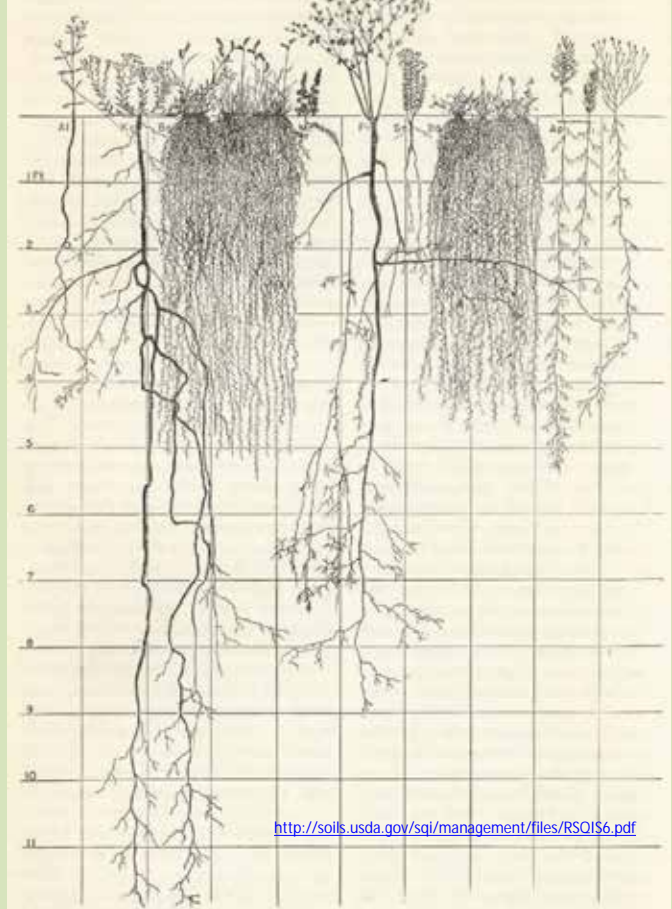


<http://www.soilandhealth.org/01aglibrary/010137veg.roots/010137ch2.html>



<http://www.bae.ncsu.edu/programs/extension/evans/ag452-1.html>

Mixed Native Perennial



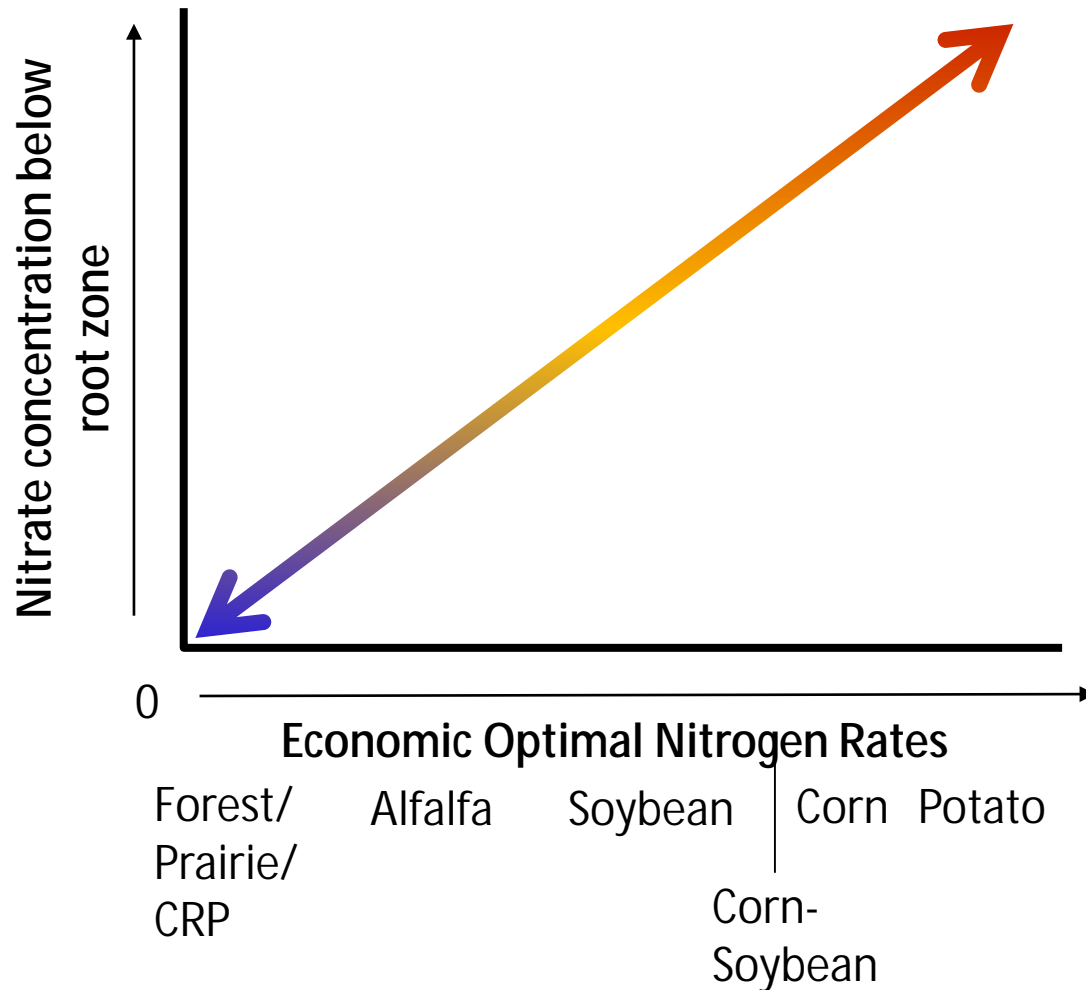
<http://soils.usda.gov/sqi/management/files/RSQIS6.pdf>

Effect of cropping systems on nitrate leaching loss in the Midwest

	Cropping systems	N Inputs kg N ha ⁻¹ yr ⁻¹	Nitrate-N Leaching kg N ha ⁻¹ yr ⁻¹	Water Drainage mm yr ⁻¹	Data Source
Annual	Corn-Corn	138	55	193	Randall et al., 1997 (1)
		180	37	399	Masarik et al., 2014 (2)
		151-221	17-32	63-187	Thomas et al., 2014 (3)
		202	63	590	Weed and Kanwar, 1996 (4)
		202	43	280	Randall and Iragavarapu, 1995 (5)
	Corn-Soybean	136-0	51	226	Randall et al., 1997 (1)
		168-0	34-46	ND	Mclsaac et al., 2010 (6)
		168-0	34	470	Weed and Kanwar, 1996 (4)
		171-0	10-35	ND	Cambardella et al., 2015 (7)
Mixed	C-S-O/A-A	171-0-57-0	8-18	ND	Cambardella et al., 2015 (7)
Perennial	Alfalfa	0	2	104	Randall et al., 1997 (1)
	CRP	0	1	160	Randall et al., 1997 (1)
	Switchgrass	0	<1-4	ND	Mclsaac et al., 2010 (6)
		112	2-11	52-156	Thomas et al., 2014 (3)
	Miscanthus	0	2-7	ND	Mclsaac et al., 2010 (6)
		112	<1-1	52-147	Thomas et al., 2014 (3)
	Prairie Pasture	0	<1	122	Masarik, et al., 2014 (2)
		0	1-10	ND	Cambardella et al., 2015 (7)

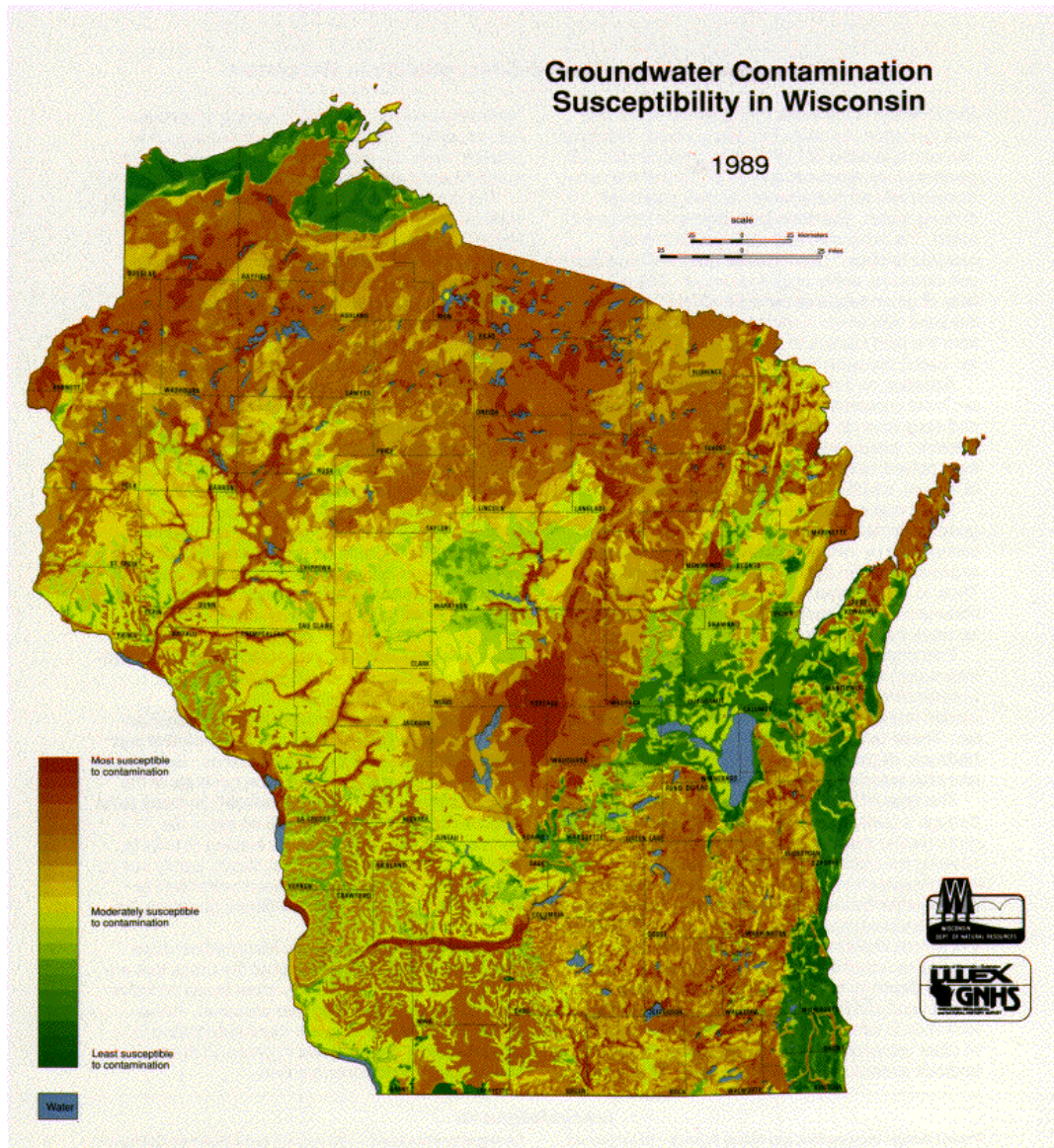
**16 -37X greater nitrate loss below continual corn cropping systems compared to perennial systems*

Nitrate Leaching Potential



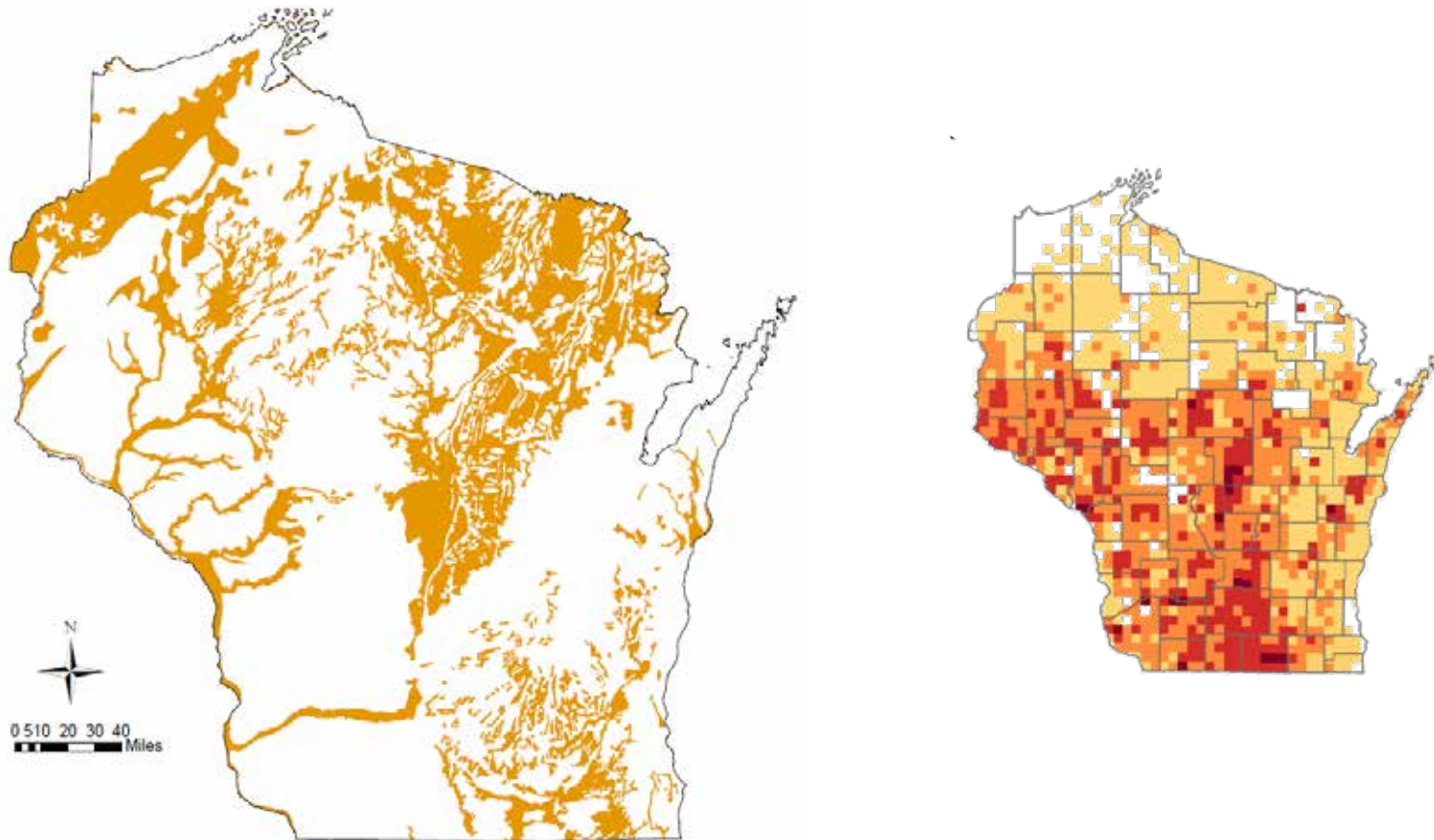
Masarik, UW-Extension

Groundwater Susceptibility



The GCSM was developed by the DNR, the US Geological Survey (USGS), the Wisconsin Geological & Natural History Survey (WGNHS), and the University of Wisconsin – Madison in the mid-1980s.

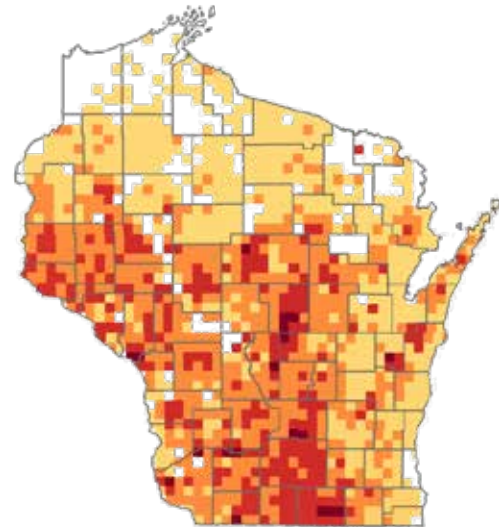
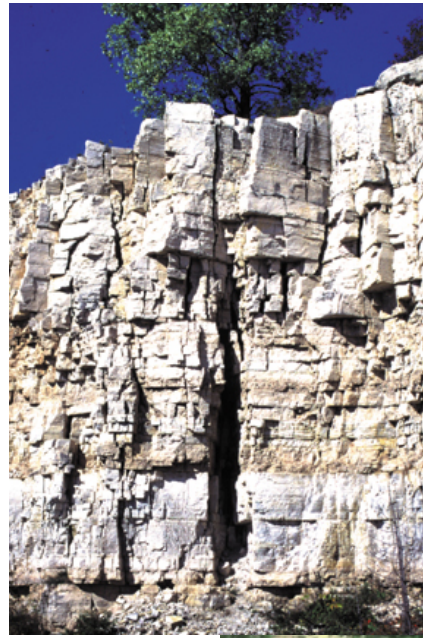
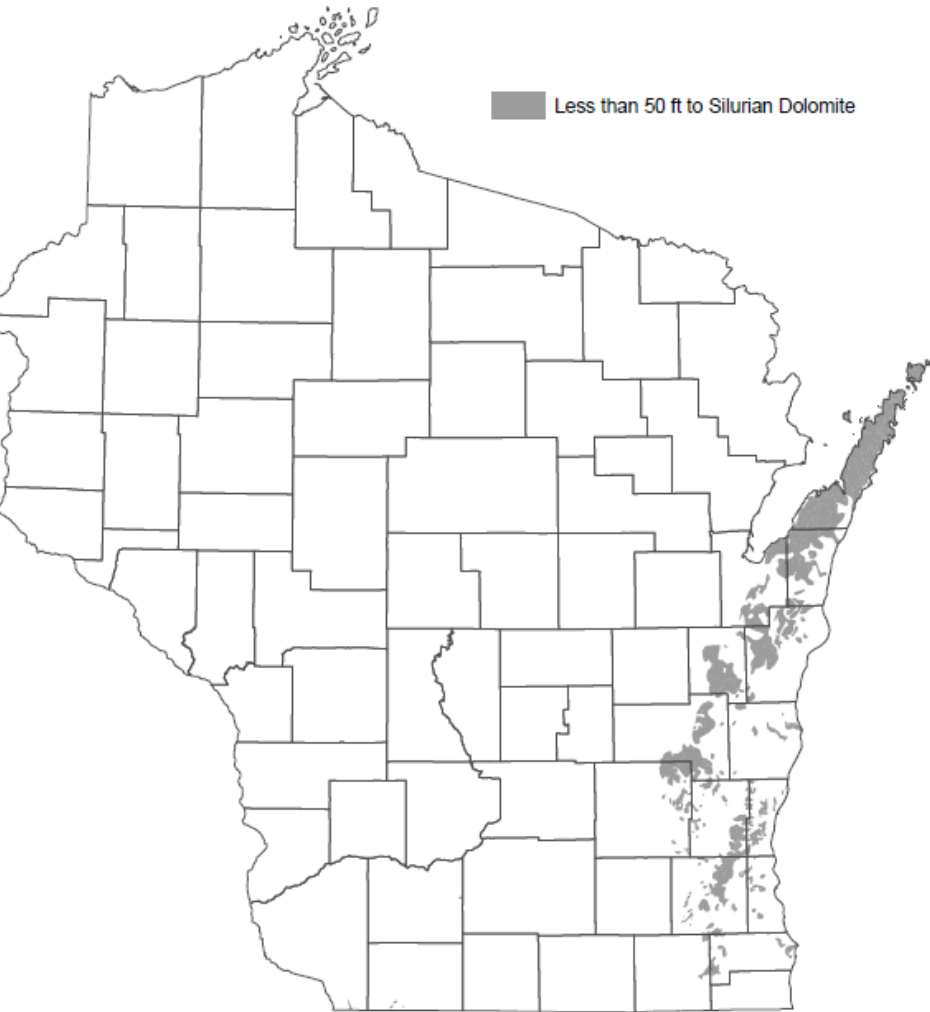
Coarse textured surficial deposits



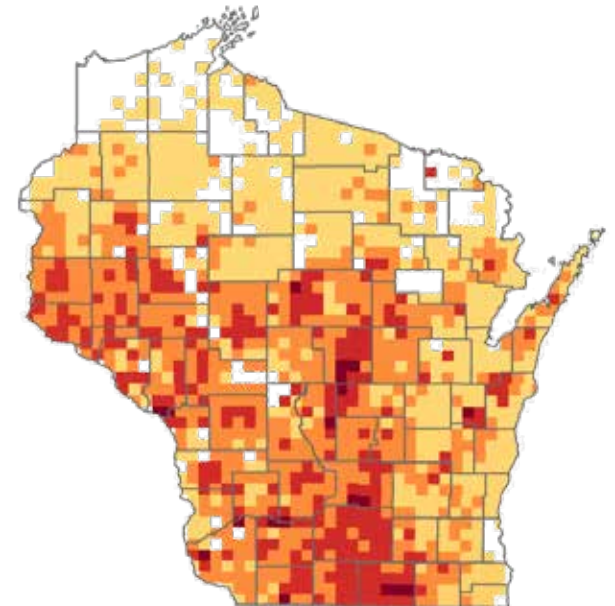
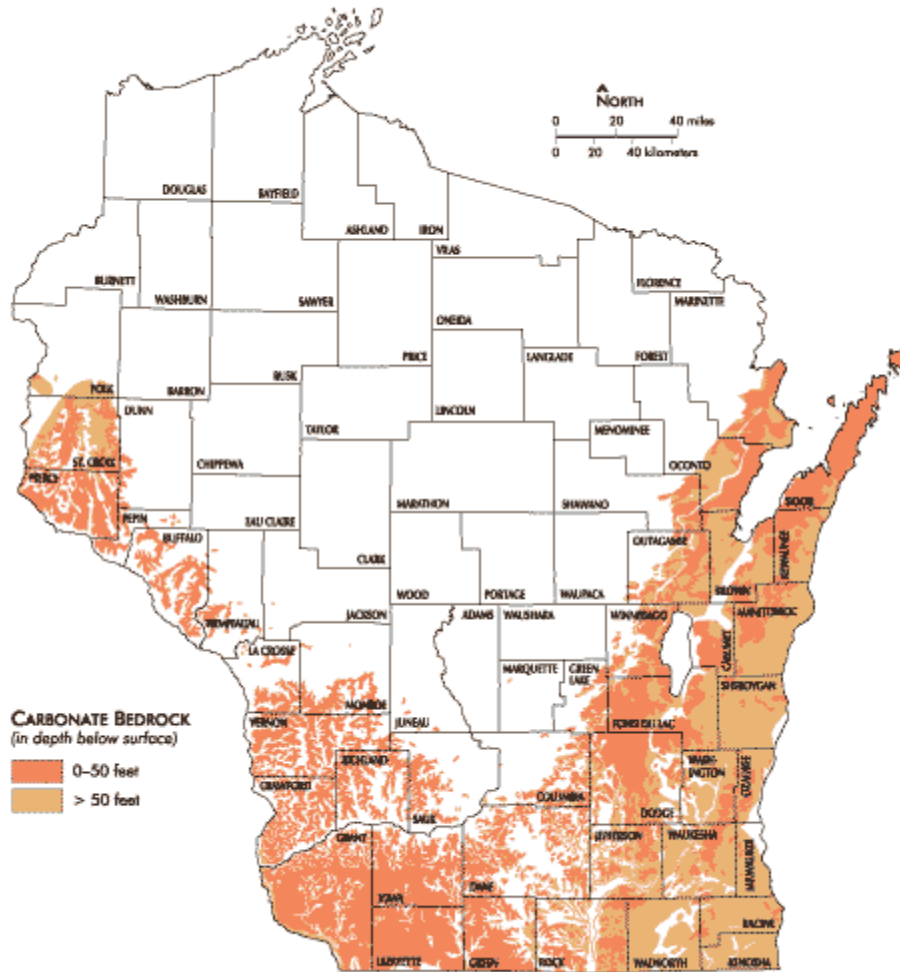
Map created using: Groundwater Contamination Susceptibility Model (GCSM); Surficial Deposits ("sdppw95c")

The GCSM was developed by the DNR, the US Geological Survey (USGS), the Wisconsin Geological & Natural History Survey (WGNHS), and the University of Wisconsin – Madison in the mid-1980s.

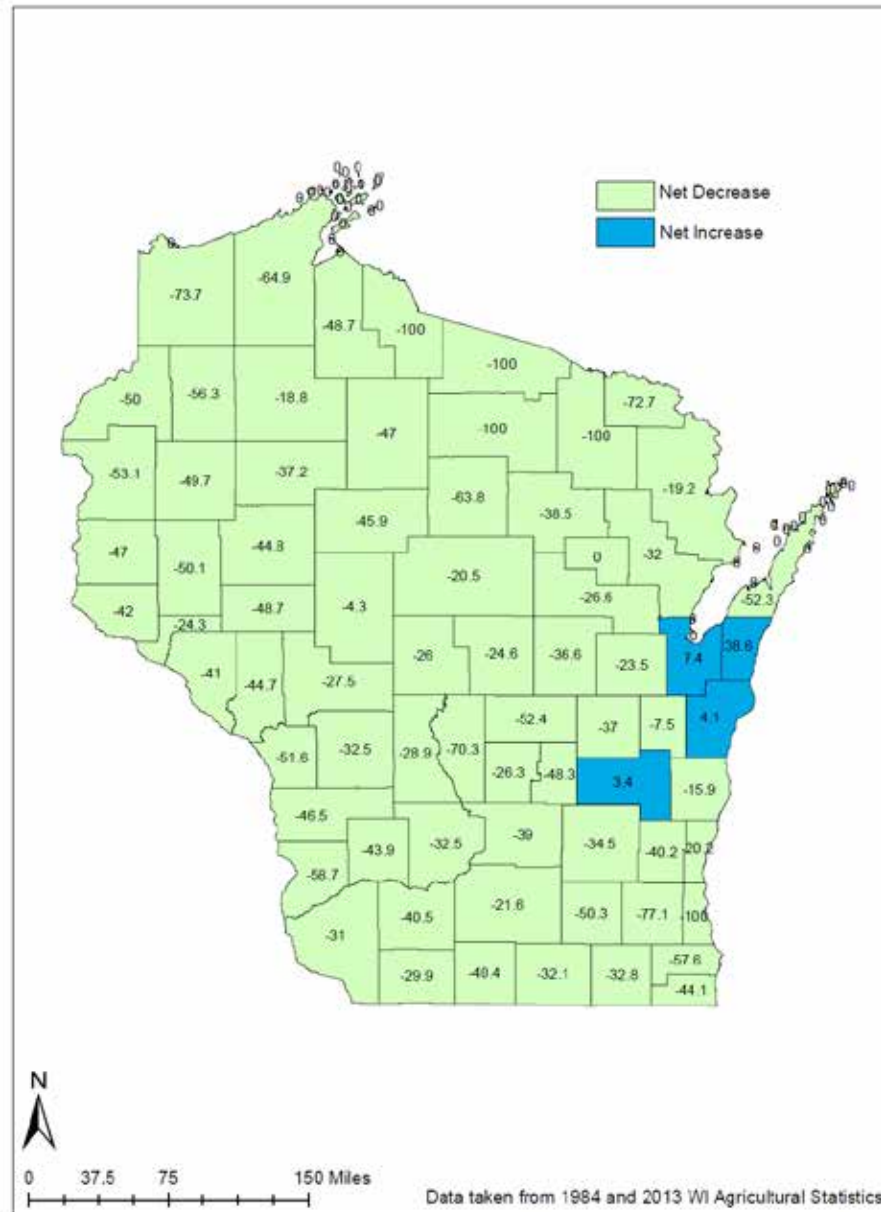
Shallow carbonate rock aquifers



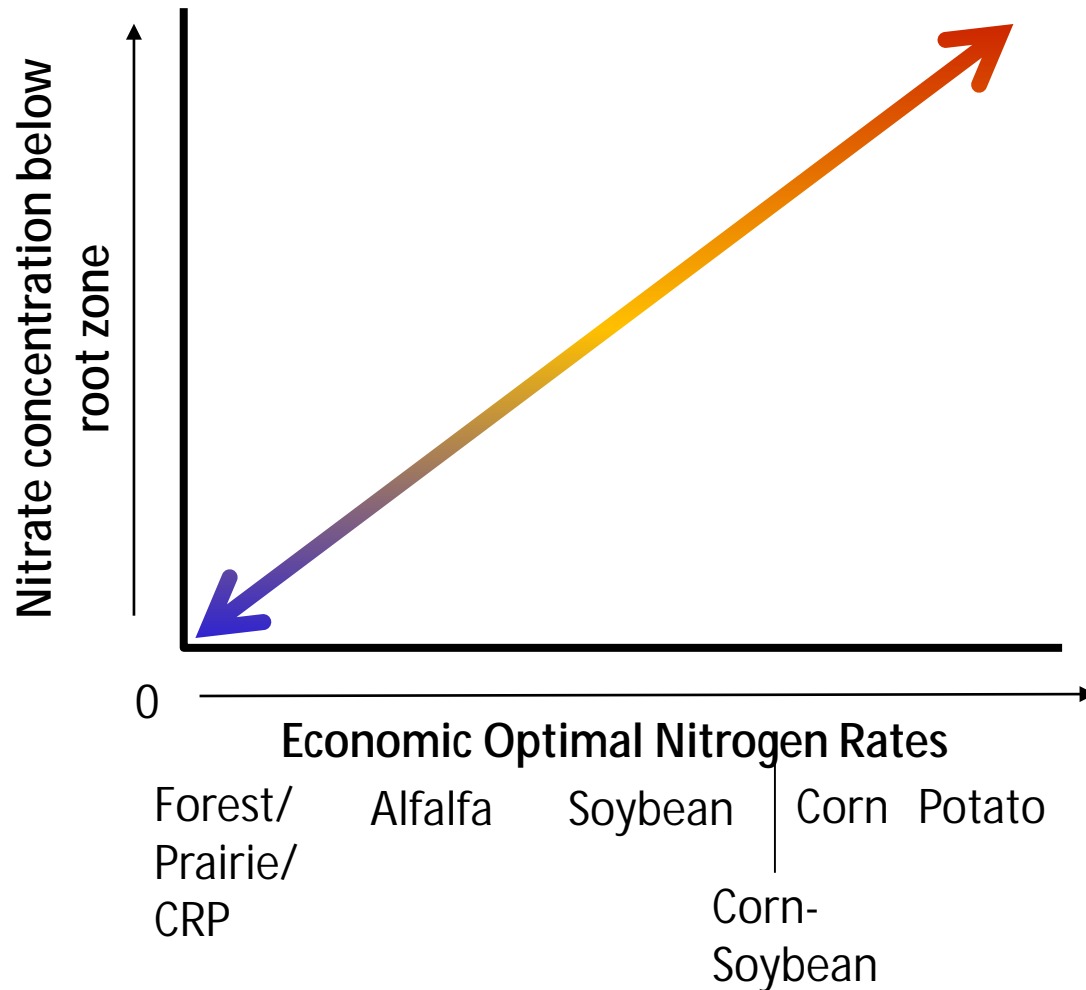
Karst Potential



% Change in Dairy Cow Numbers from 1983-2012 by County

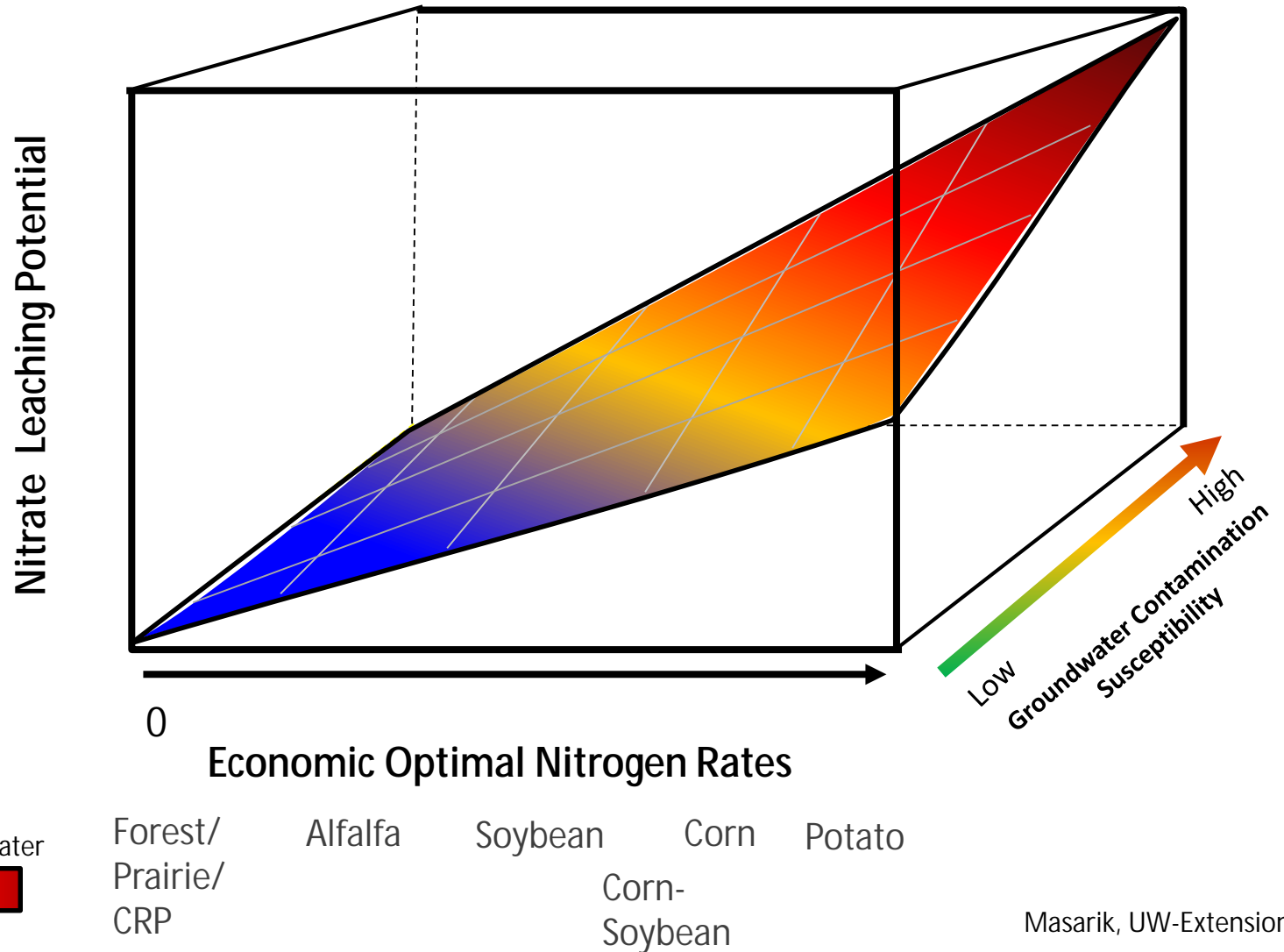


Nitrate Leaching Potential

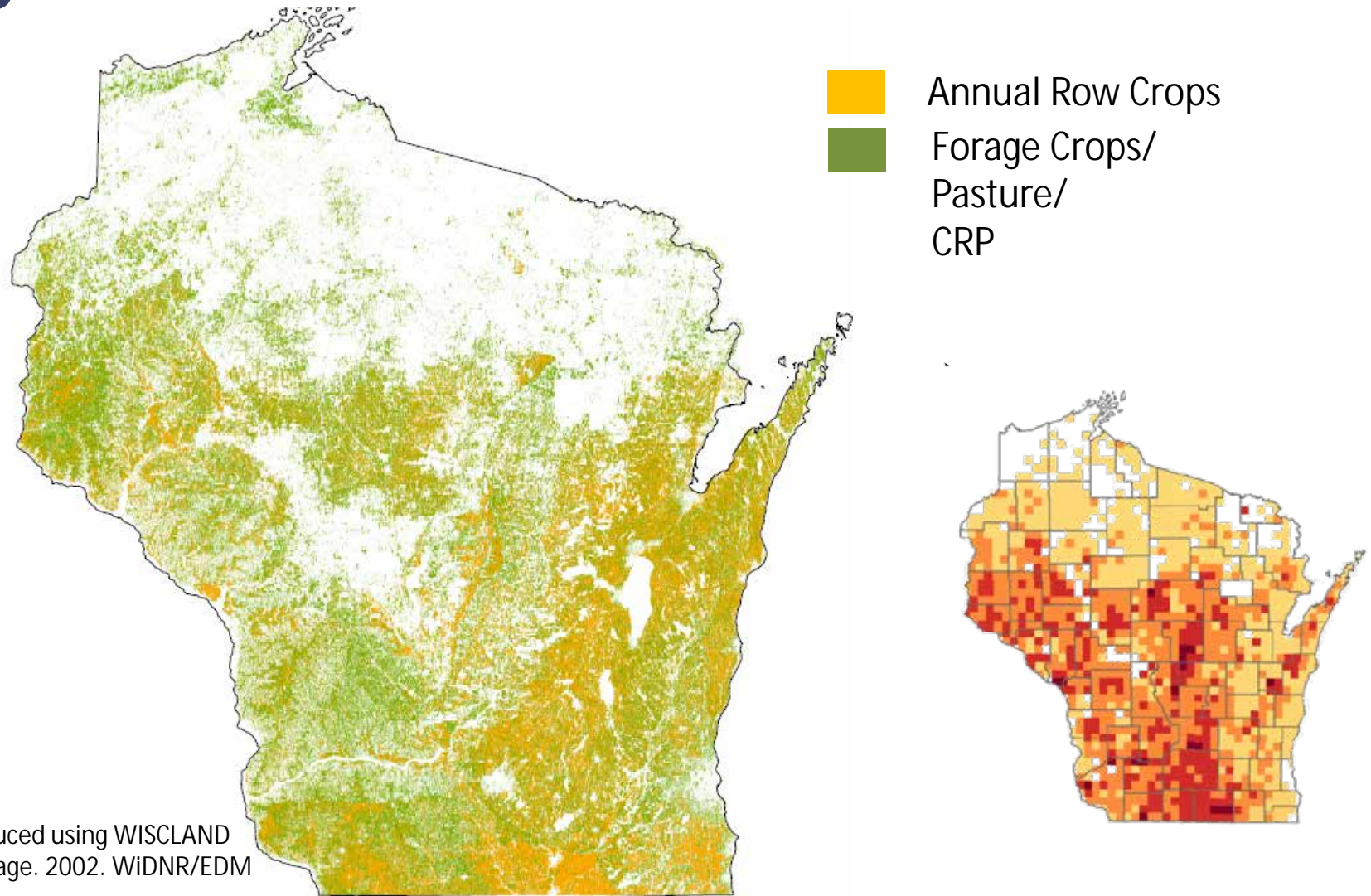


Masarik, UW-Extension

Nitrate Leaching Potential



Agricultural Lands of Wisconsin



Agricultural Lands of Wisconsin



Row Cropping Systems



Forage Crops/Pasture/CRP

Maps produced using WISCLAND Data Coverage. 2002. WIDNR/EDM



2013 Area of Interest



Land Cover Categories
(by decreasing acreage)

AGRICULTURE*

- Grass/Pasture
- Corn
- Soybeans
- Alfalfa
- Winter Wheat
- Other Hay/Non Alfalfa
- Oats
- Sweet Corn
- Dbt Crop Win/Wht/Corn
- Rye
- Fallow/Idle Cropland
- Sorghum
- Peas
- Dbt Crop Win/Wht/Soybeans
- Sod/Grass Seed
- Dbt Crop Soybeans/Oats

NON-AGRICULTURE**

- Deciduous Forest
- Developed/Open Space
- Developed/Low Intensity
- Herbaceous Wetlands
- Woody Wetlands
- Developed/Medium Intensity





2014 Area of Interest



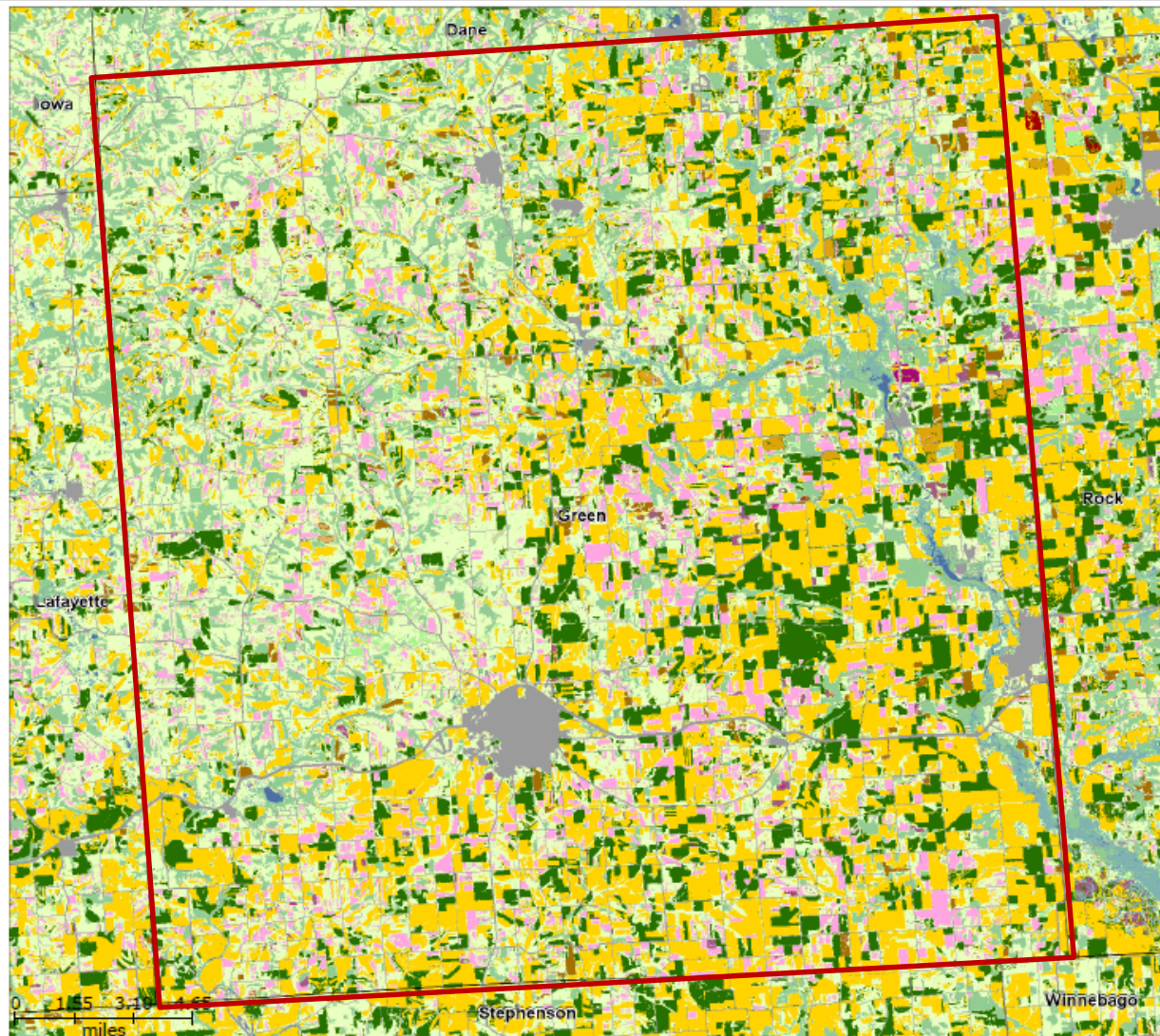
Land Cover Categories
(by decreasing acreage)

AGRICULTURE*

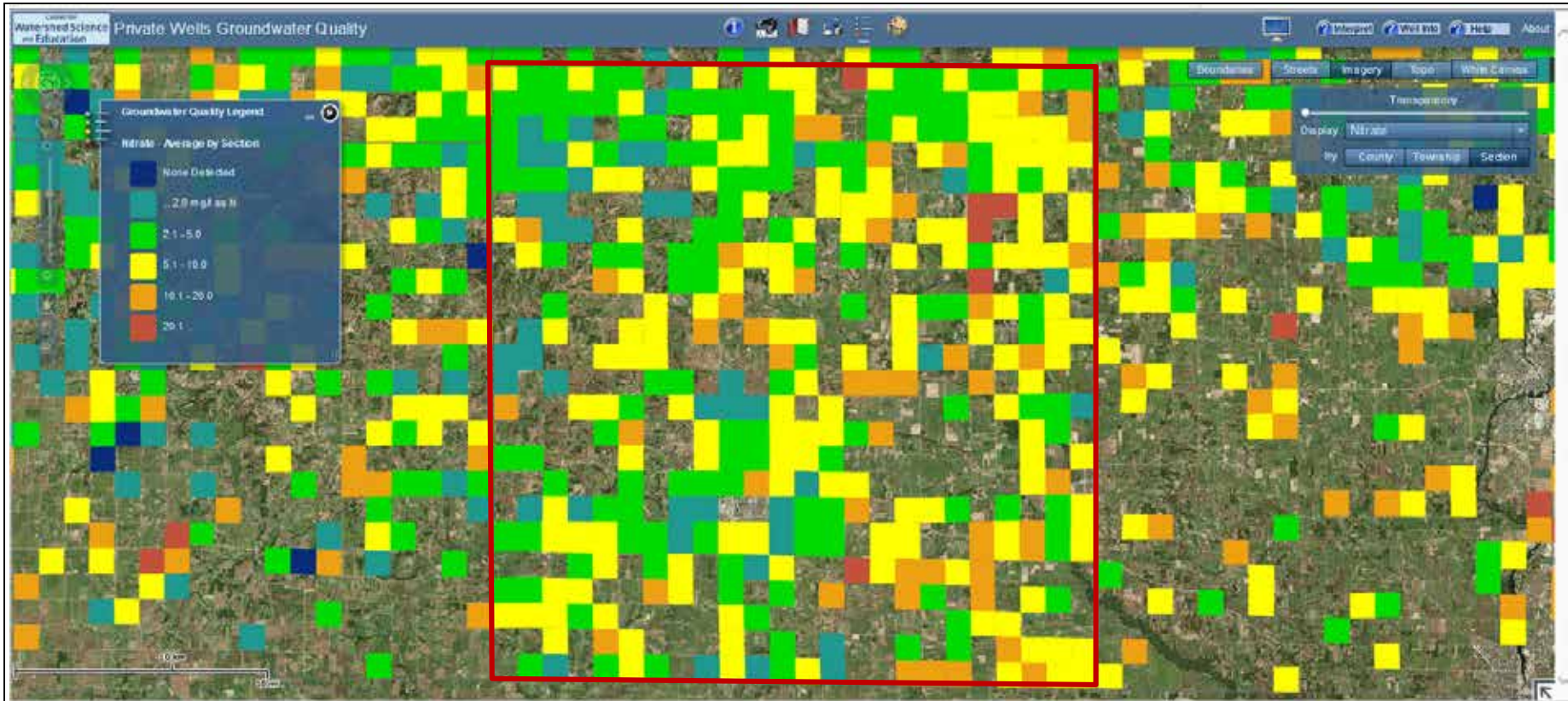
- Grass/Pasture
- Corn
- Soybeans
- Alfalfa
- Winter Wheat
- Other Hay/Non Alfalfa
- Oats
- Sweet Corn
- Dbl Crop WinWht/Corn
- Rye
- Dry Beans
- Fallow/Idle Cropland
- Spring Wheat
- Barley
- Dbl Crop WinWht/Soybeans
- Triticale

NON-AGRICULTURE**

- Deciduous Forest
- Developed/Open Space
- Developed/Low Intensity
- Herbaceous Wetlands
- Woody Wetlands
- Developed/Medium Intensity



Average Nitrate-N concentration by section.



Range	Number	Percent	Summary
None Detected	194	7%	Minimum: No Detect
... 2.0	622	23%	
2.1 - 5.0	724	27%	Median: 4.1
5.1 - 10.0	694	26%	Average: 5.61288
10.1 - 20.0	364	14%	
20.1 ...	64	2%	Maximum: 69.9
Total	2662		
> 10mg/l N	428	16%	Exceeds Health Standard

<http://www.uwsp.edu/cnr-ap/watershed/Pages/wellwaterviewer.aspx>

Green County
Nitrate Summary

Factors affecting nitrogen loss to groundwater

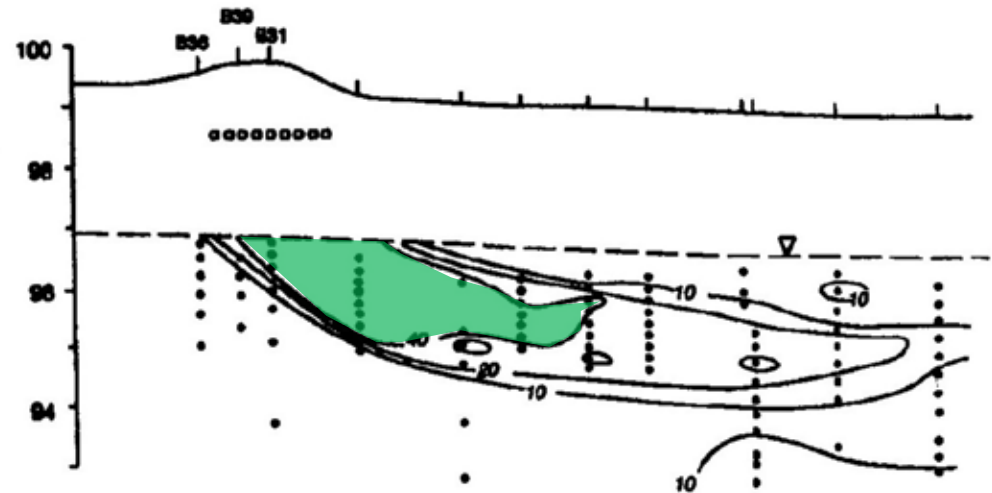
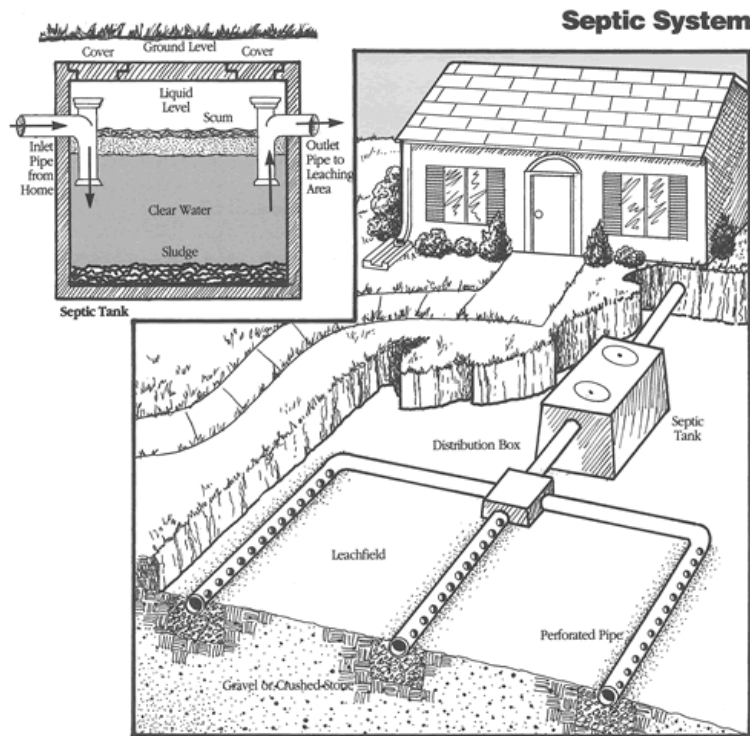
Within our control

- Amount of nitrogen applied
 - As a function of cropping system
 - Nitrogen application rate relative to economic optimum – *right amount*
 - When, where, what form
- Percent of land base in production

Out of our control

- Geology
- Soil Type
- Precipitation / Climate

Septic systems and nitrate



Robertson and Harman 1999

- Designed to dispose of human waste in a manner that prevents bacteriological contamination of groundwater supplies.
- **Do not** effectively remove all contaminants from wastewater:
Nitrate, chloride, *viruses?*, *pharmaceuticals?*, *hormones?*

Comparing Land-use Impacts



	Corn ¹ (per acre)	Prairie ¹ (per acre)	Septic ² System
Total Nitrogen Inputs (lb)	169	9	20-25
Nitrogen Leaching Loss (lb)	32	0.04	16-20
Amount N lost to leaching (%)	19	0.4	80-90

1 Data from Masarik, 2014

2 Data from Tri-State Water Quality Council, 2005 and EPA 625/R-00/008

Comparing Land-use Impacts



20 acres

32 lbs	32 lbs	32 lbs	32 lbs
32 lbs	32 lbs	32 lbs	32 lbs
32 lbs	32 lbs	32 lbs	32 lbs
32 lbs	32 lbs	32 lbs	32 lbs
32 lbs	32 lbs	32 lbs	32 lbs

$32 \text{ lbs/ac} \times 20 \text{ acres} = 640 \text{ lbs}$
14 mg/L



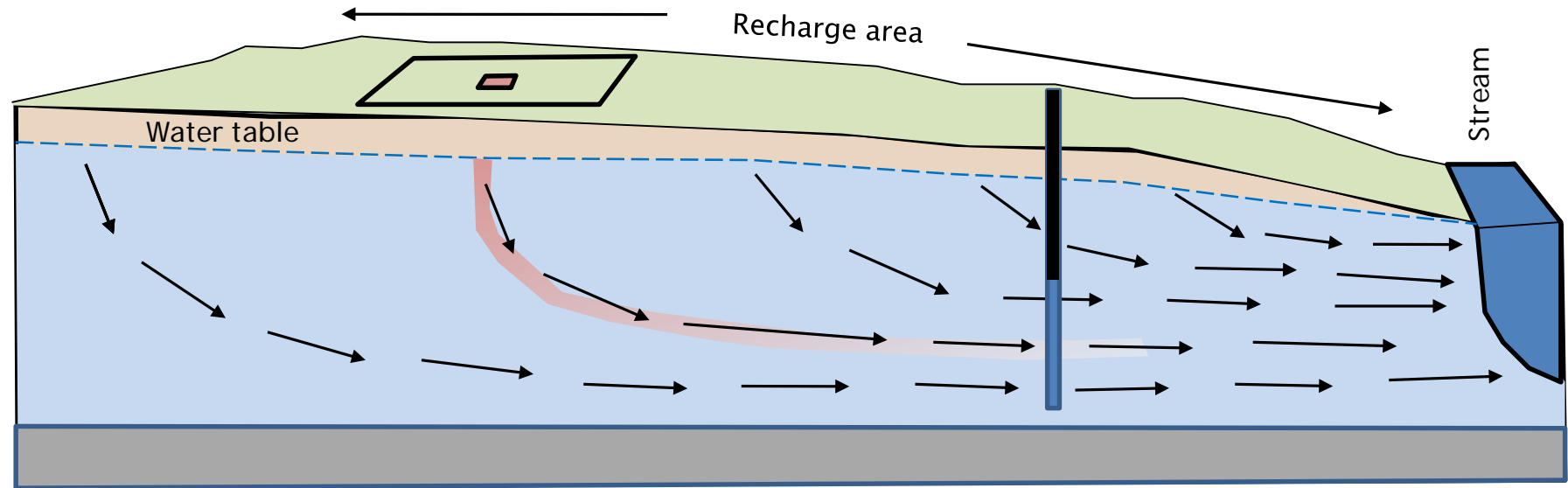
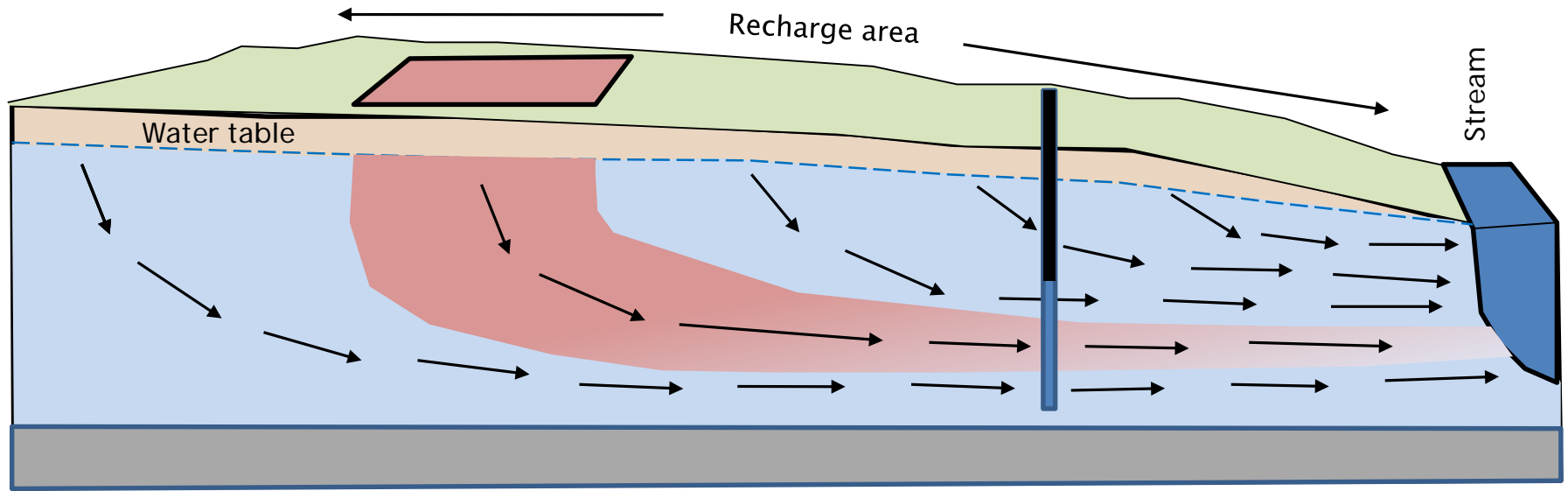
20 acres

20 lbs

$20 \text{ lbs/septic system} \times 1 \text{ septic systems} = 20 \text{ lbs}$
1/32nd the impact on water quality
0.44 mg/L

Assuming 10 inches of recharge

32 lbs/ac x 20 acres = 640 lbs



20 lbs/septic system

Comparing Land-use Impacts



20 acres

32 lbs	32 lbs	32 lbs	32 lbs
32 lbs	32 lbs	32 lbs	32 lbs
32 lbs	32 lbs	32 lbs	32 lbs
32 lbs	32 lbs	32 lbs	32 lbs
32 lbs	32 lbs	32 lbs	32 lbs
32 lbs	32 lbs	32 lbs	32 lbs

$32 \text{ lbs/ac} \times 20 \text{ acres} = 640 \text{ lbs}$

20 acres

20 lbs	20 lbs	20 lbs	20 lbs
20 lbs	20 lbs	20 lbs	20 lbs
20 lbs	20 lbs	20 lbs	20 lbs
20 lbs	20 lbs	20 lbs	20 lbs
20 lbs	20 lbs	20 lbs	20 lbs
20 lbs	20 lbs	20 lbs	20 lbs
20 lbs	20 lbs	20 lbs	20 lbs
20 lbs	20 lbs	20 lbs	20 lbs

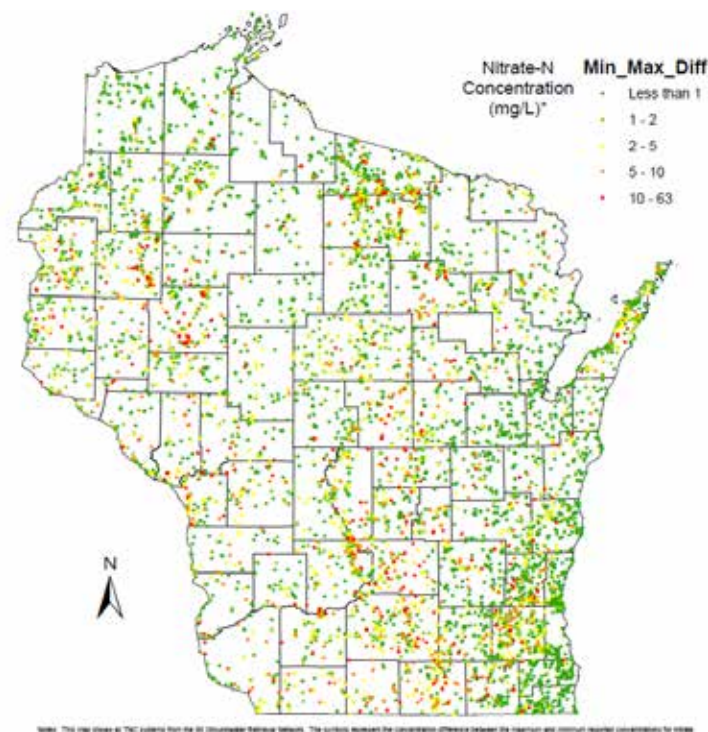
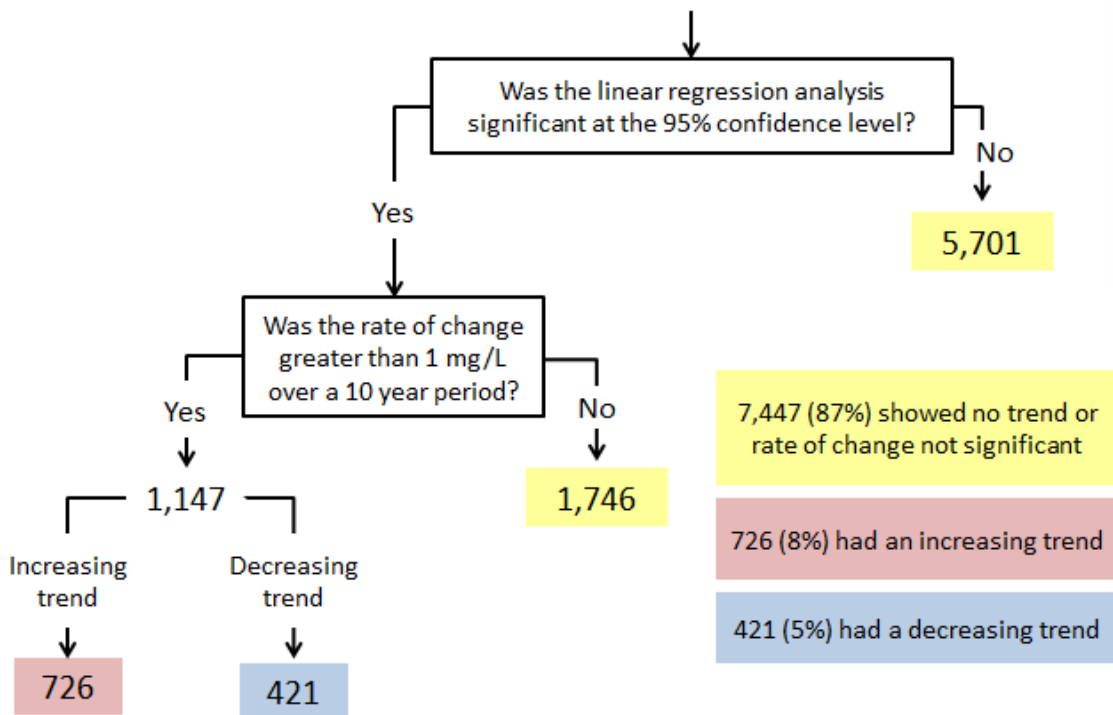
$20 \text{ lbs/septic system} \times 32 \text{ septic systems} = 640 \text{ lbs}$

Using these numbers: 32 septic systems on 20 acres (0.6 acre lots) needed to achieve same impact to water quality as 20 acres of corn



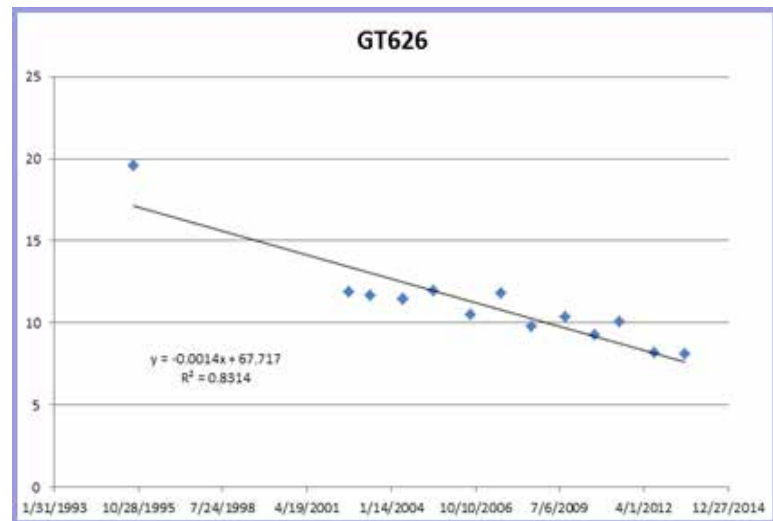
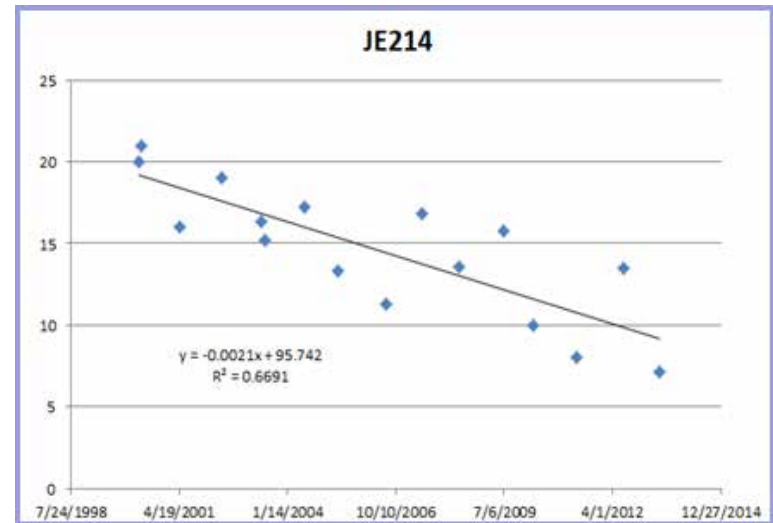
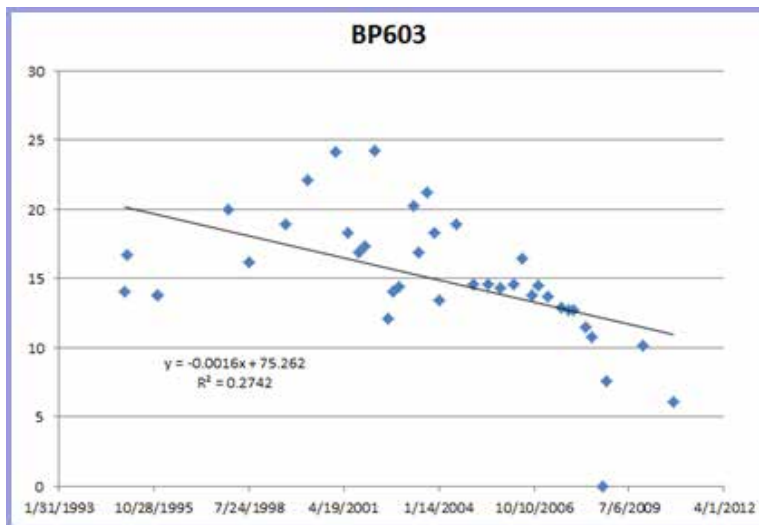
Nitrate Trends

Transient Non-Community Well Water Systems
8,594 systems

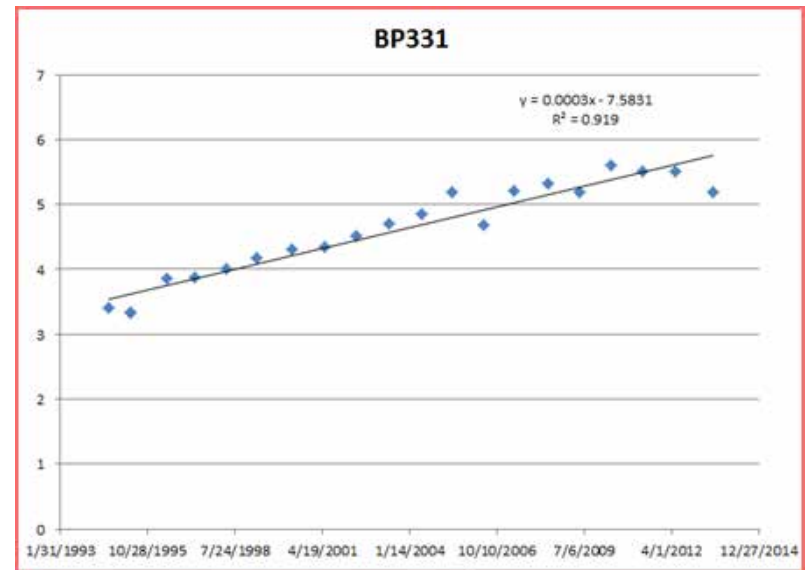
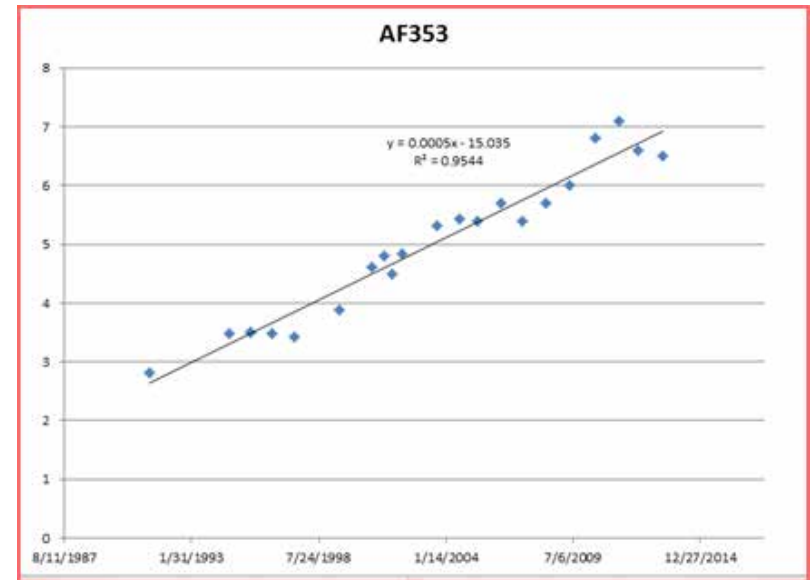
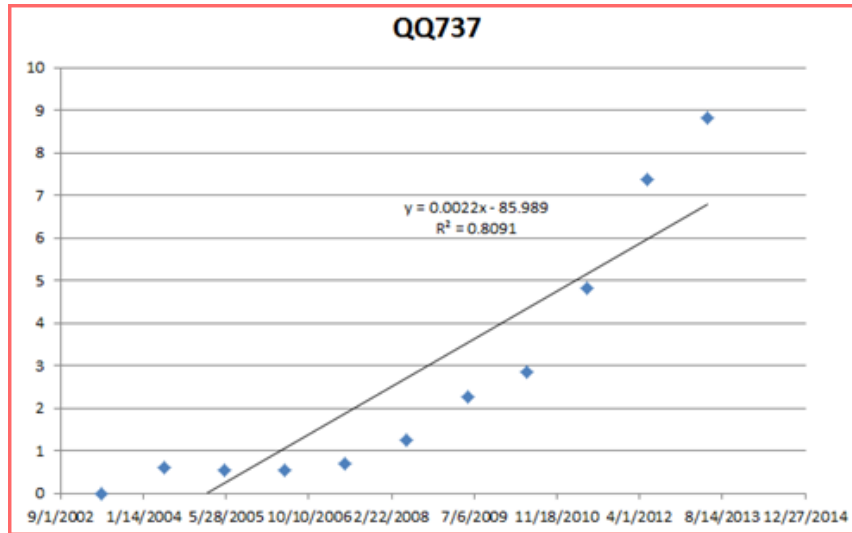


[Masarik et al., 2014](#)

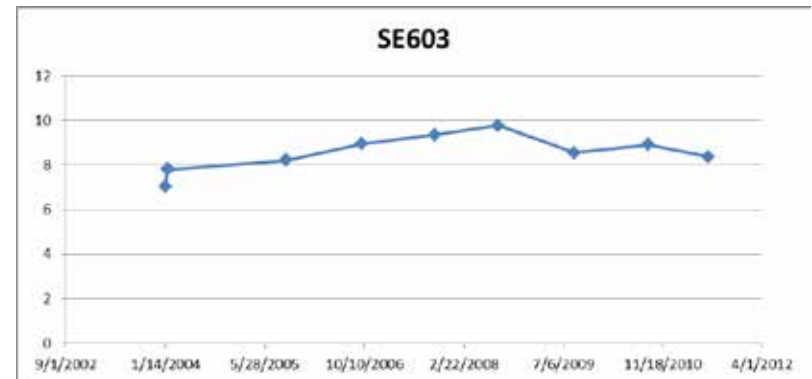
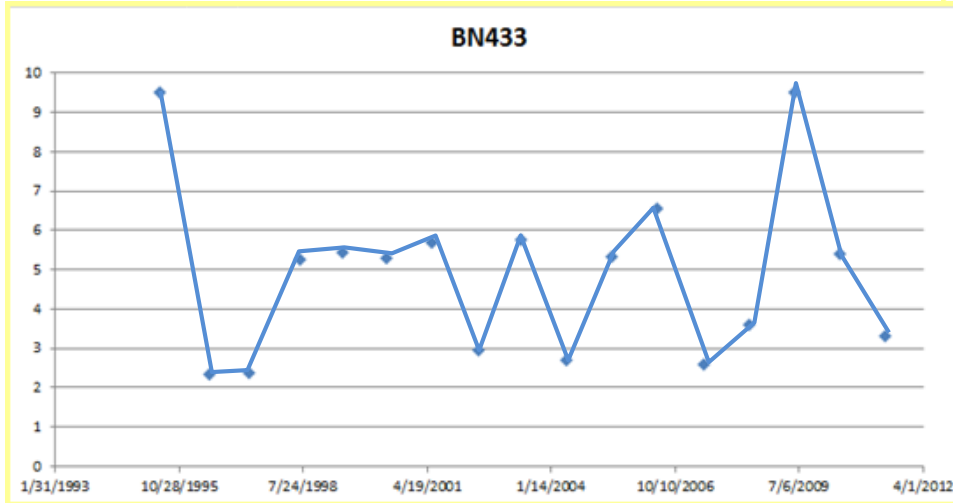
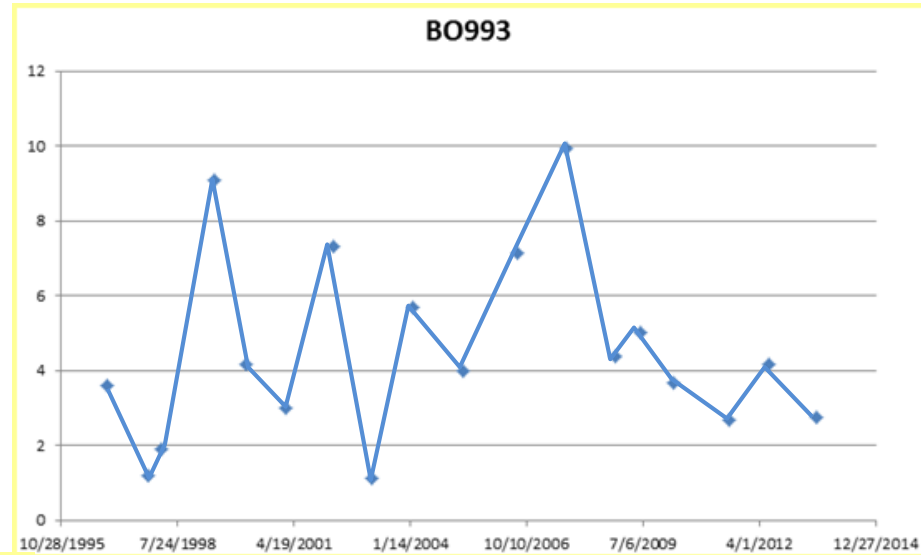
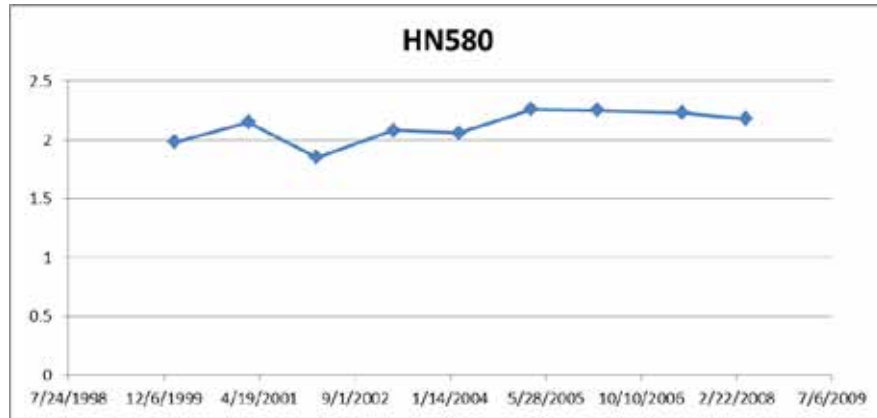
Examples of TNC wells with decreasing trend



Examples of TNC wells with an increasing trend

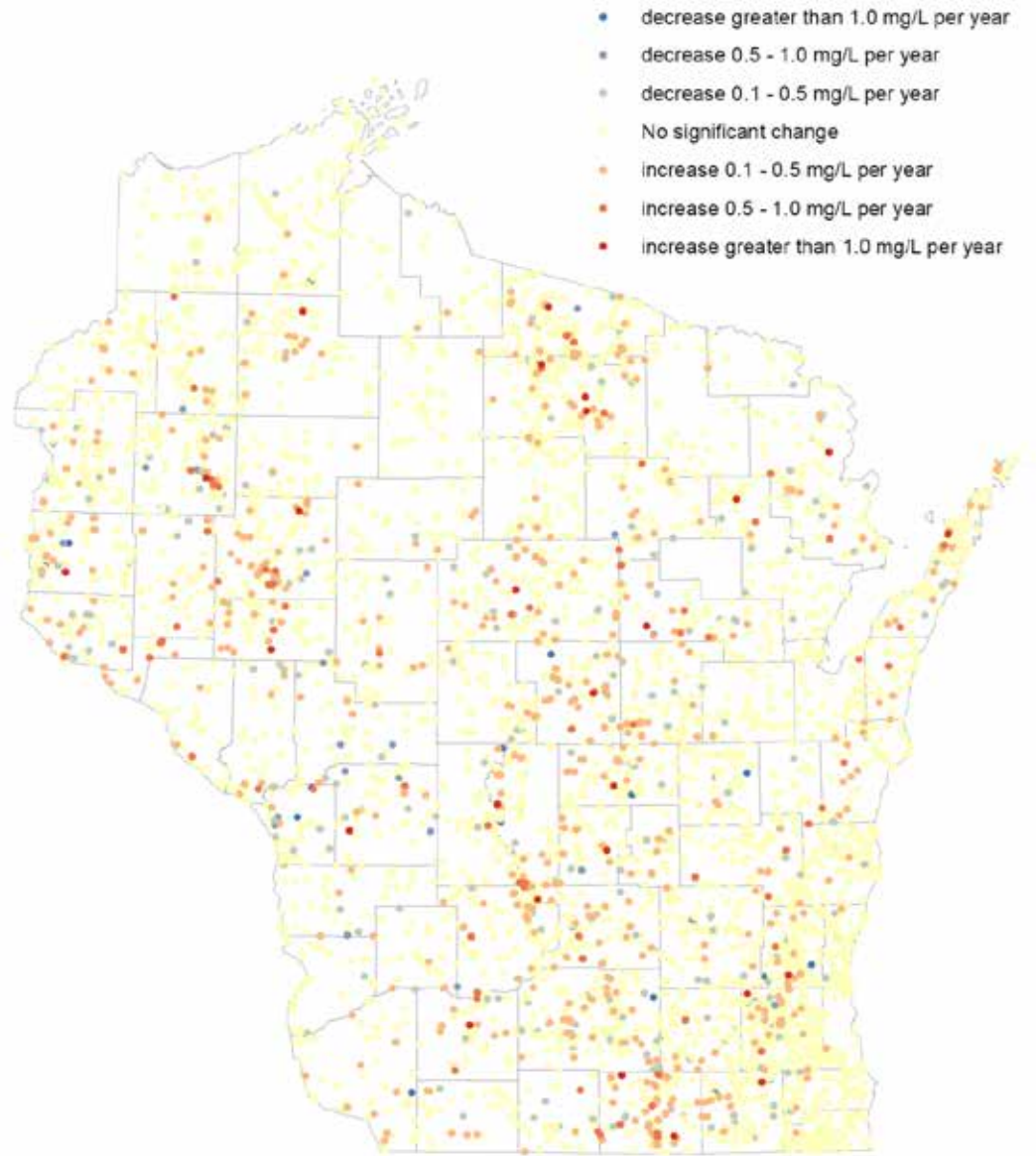


Examples of TNC wells w/no trend

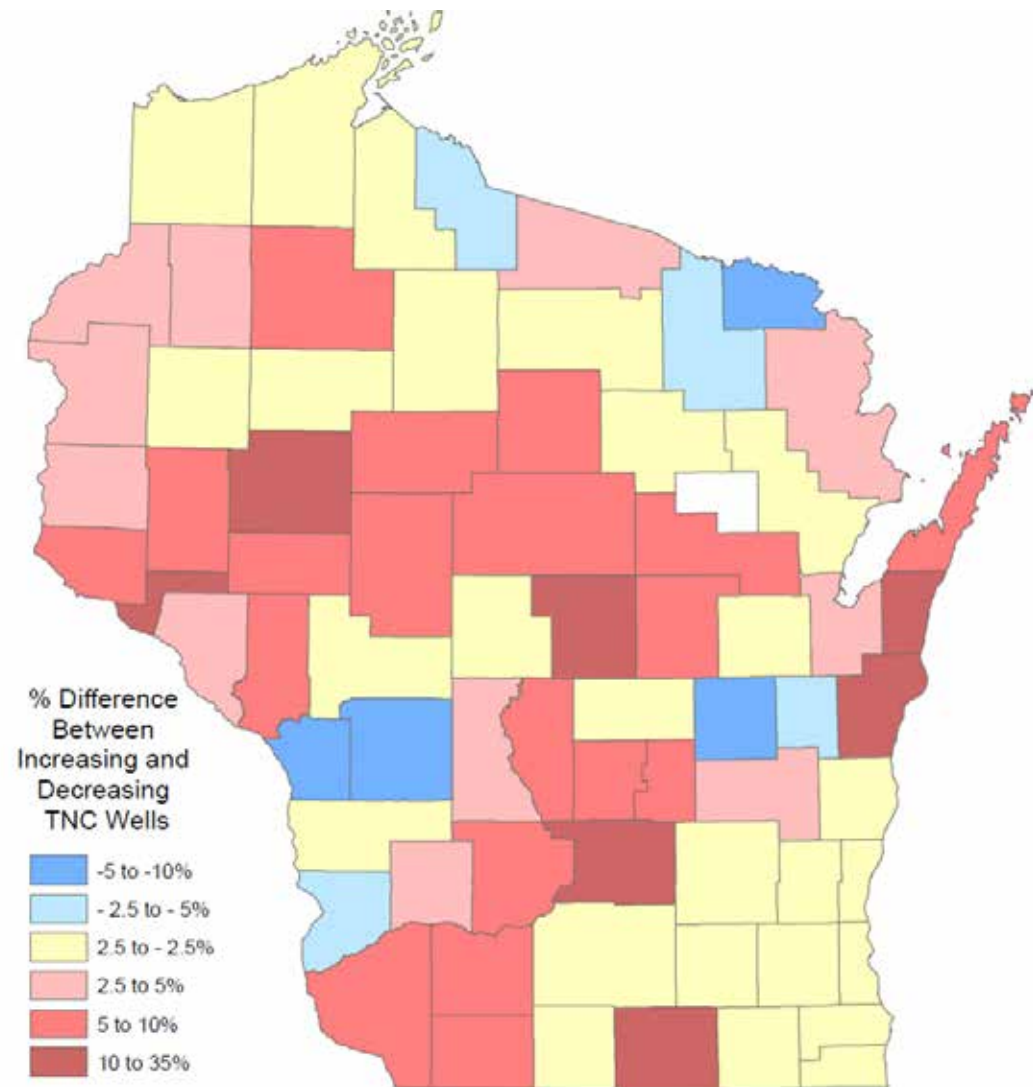


Location and result for TNC wells

- decrease greater than 1.0 mg/L per year
- decrease 0.5 - 1.0 mg/L per year
- decrease 0.1 - 0.5 mg/L per year
- No significant change
- increase 0.1 - 0.5 mg/L per year
- increase 0.5 - 1.0 mg/L per year
- increase greater than 1.0 mg/L per year

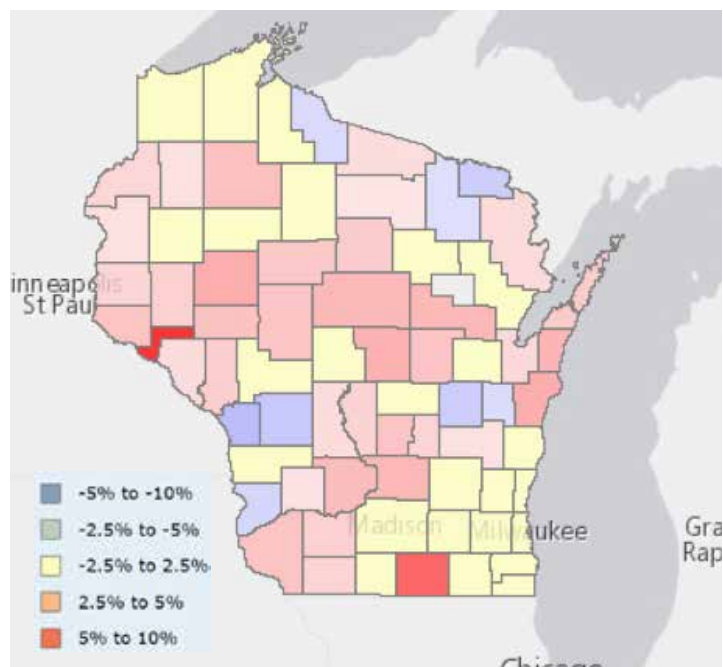


Counties that
have seen more
TNC wells
increase (red)
or
decrease (blue)

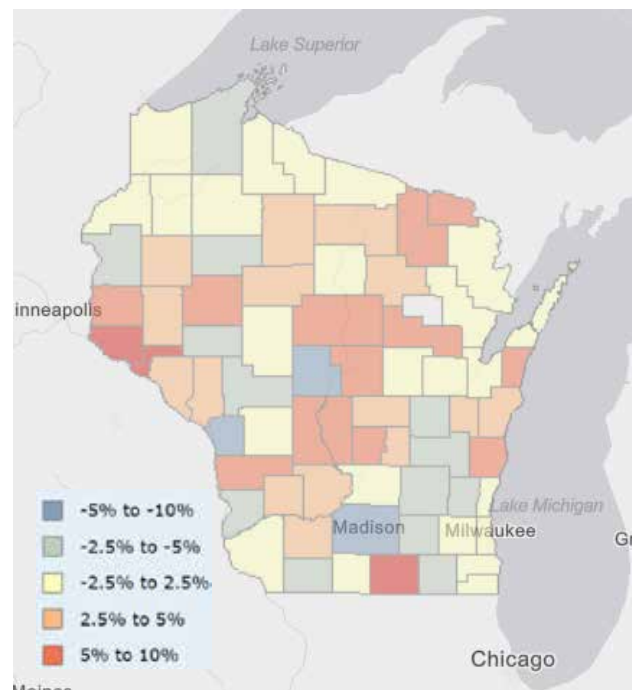


[Masarik et al., 2014](#)

Nitrate Trends by County



All TNC data, going back ~20 years



All TNC and NTNC systems limited to previous 10 years of data.

<http://dnr.wi.gov/topic/groundwater/GCC/gwquality.html>

Long-term nitrogen reduction strategies

Practice	Details	% Nitrate-N Reduction (SD)
Timing	Fall to Spring Pre-plant	6 (25)
	Spring pre-plant/sidedress 40-60 split compared to fall applied	5 (28)
	Sidedress – Soil test based compared to pre-plant	7 (37)
Nitrification Inhibitor	Nitrapyrin – Fall – Compared to applied w/out nitrapyrin	9 (19)
Cover Crops	Rye	31 (29)
	Oat	28 (2)
Perennial	Biofuel Crops (ex. switchgrass, miscanthus)	72 (23)
	Conservation Reserve Program	85 (9)
Extended Rotations	At least 2 years of alfalfa or other perennial crops in a 4 or 5 year rotation	42 (12)

[Iowa Nutrient Reduction Strategy, 2014](#)

How Manure Composition Affects N Mineralization

The rate of mineralization in soil depends upon the "digestibility" of manure organic matter and its carbon:nitrogen (C:N) ratio. Separation of whole manure into liquids and solids segregates coarse and fine manure particles that have different organic composition and different mineralization rates. Fine particles in manure contain organic compounds with low C:N ratios (high protein) and are rapidly decomposed in soil. Coarse particles have higher C:N ratios (lower protein) and are **more slowly decomposed in soil.**

Because thin slurry and lagoon water contain the finest organic particles, these materials have the most rapid N mineralization rate. Thick slurry and solid manures contain a mixture of fine and coarse particles, so they have a lower N mineralization rate.

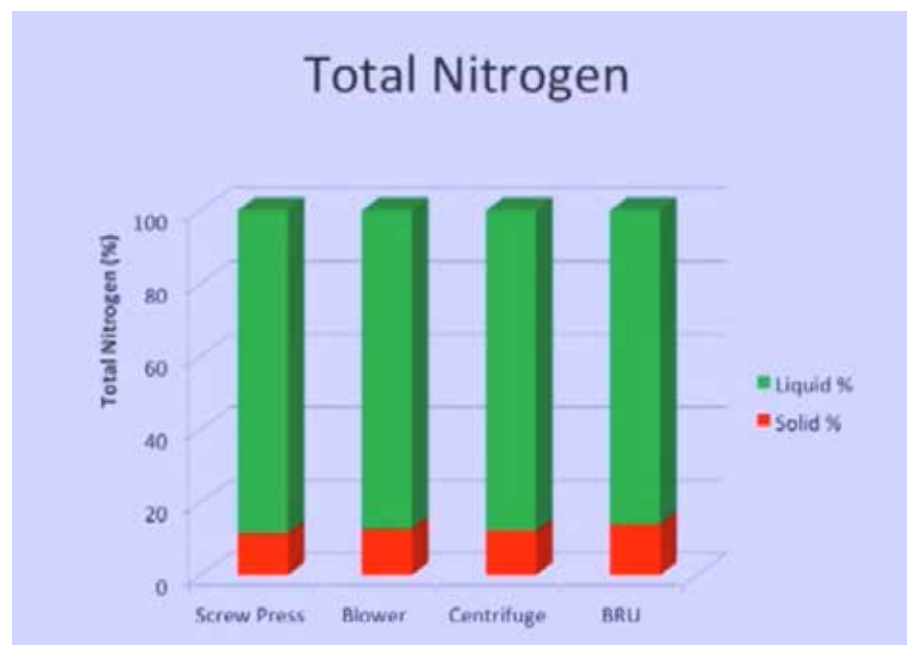
Solids separated from liquid manure by a mechanical separator (separated dairy solids) contain mostly coarse particles (bedding plus undigested feed). These solids have a unique pattern of mineralization over time in soil. Separated solids typically have negative N mineralization rates (PAN in soil decreases) for 4 to 8 weeks after application. After that, PAN is mineralized very slowly. Cumulative PAN from separated solids is much lower than for other fresh manures. The timing and amount of PAN release from horse manure is similar to that from separated dairy solids.

Separation of solids from liquid manure by gravity separation (settling basin or evaporation basin) does not change PAN, because the fine organic particles in the manure are recovered from the basin.

Composting manure reduces manure volume by 50 percent or more. During composting, some of the manure N is lost as ammonia gas, and some is transformed to more stable organic compounds. Compost organic matter decomposes very slowly in soil. Cumulative PAN for compost organic matter is similar to that of separated dairy solids.

Fresh poultry manure or broiler litter contains some organic N in the form of uric acid (similar to urea). In soil, uric acid is converted to PAN in 1 to 2 weeks. Most broiler litter sold as "compost" in western Oregon contains uric acid and behaves more like fresh litter than compost in terms of N availability. If you can smell ammonia in broiler litter, it probably is not thoroughly composted. Dry-stacking of broiler litter does not provide adequate moisture for composting.

Because thin slurry and lagoon water contain the finest organic particles, these materials have the most rapid N mineralization rate. Thick slurry and solid manures contain a mixture of fine and coarse particles, so they have a lower N mineralization rate.



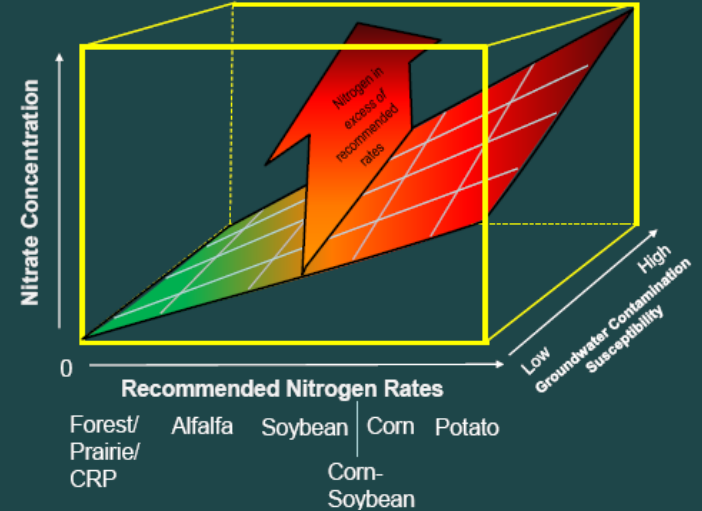
Slide from presentation by Becky Larson, Manure Irrigation Workshop

Liquid contains greater percentage of plant available N (i.e. ammonia/ammonium), ammonia readily converted into nitrate in aerobic conditions and susceptible to leaching.

Conclusions

- Nutrient management is a first step that creates a baseline concentration of nitrate in groundwater that reflects crop rotation and geology/soils.
- Significant nitrate leaching can occur even when nitrogen recommendations are followed – no environmental optimum rate
- Nutrient management and crediting of N will help reduce extreme nitrate concentrations in groundwater and reduce risk of brown water incidents in groundwater
- May take years or decades for groundwater quality to reflect changes in land-use practices

$$\text{Nitrate} = f(\text{Crop} + \text{Excess N} + \text{soils/geology})$$



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What can be done to reduce nitrate levels?

q Short term

q Municipal Wells ([GCC, 2015](#))

- q 47 systems have spent >\$32 million as of 2012
 - q Water Treatment
 - q New wells
 - q Blending

q Private Wells ([Lewandowski et. al. 2008](#))

- q New well (not guaranteed, deeper adds to expense) - \$7,200
- q Bottled water - \$190/person/year
- q Water treatment devices \$800 + 100/yr
 - q Reverse osmosis
 - q Distillation
 - q Anion exchange

YouTube video player showing a sand-tank groundwater model. The video title is "Aquifer penetration video using sand-tank groundwater model" by Kevin Masarik. The video shows a cross-section of a sand tank with green dye being injected into the sand, illustrating the flow of water and the resulting dye plume. The video is 0:01 / 0:18 long.

Aquifer penetration video using sand-tank groundwater model
Kevin Masarik · 2 videos · 261 views

Published on Mar 12, 2013
Green dye is used to illustrate aquifer penetration. There is a lag between what happens on the land surface and the subsequent water quality within the aquifer, as well as water quality in the river. [Show more](#)

ALL COMMENTS (1)

Share your thoughts

Top comments

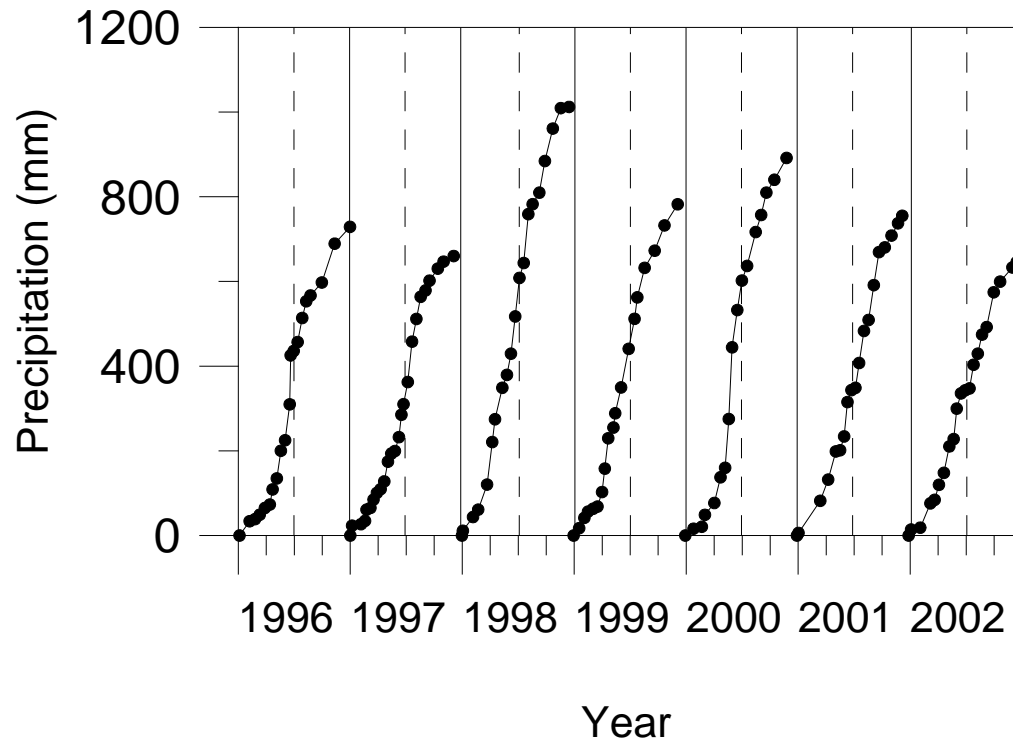
Kevin Masarik · 11 months ago

Recommended videos:

- Dowsing for Gold.wmv by Reed Luhrs · 88,388 views · 29:52
- Groundwater by Christopher Hewitt · 556 views · 22:37
- Dowsing II with Robert Rohe 6-10-10 by storytelus · 19,328 views · 9:53
- Edwards Aquifer by snowvideo · 258 views · 7:04
- MATC Earth Science- Groundwater Model Demo by Scott Johnson · 1,245 views · 6:32
- Hydrology Basics and the Northern Arizona Regional Groundwater Flow by Verde River Basin Partnership · 781 views · 28:09
- Cave Diving Floridan Aquifer by Rivercast · 9,310 views · 13:28
- What is groundwater? by Waterscapes · 6,127 views · 16:10
- RET by lov2hike · 143 videos · 15:00
- Ground Water Model by P3Upested · 12 views · 15:00

<http://www.youtube.com/watch?v=BKrN2HdvGp4>

Annual Cumulative Precipitation

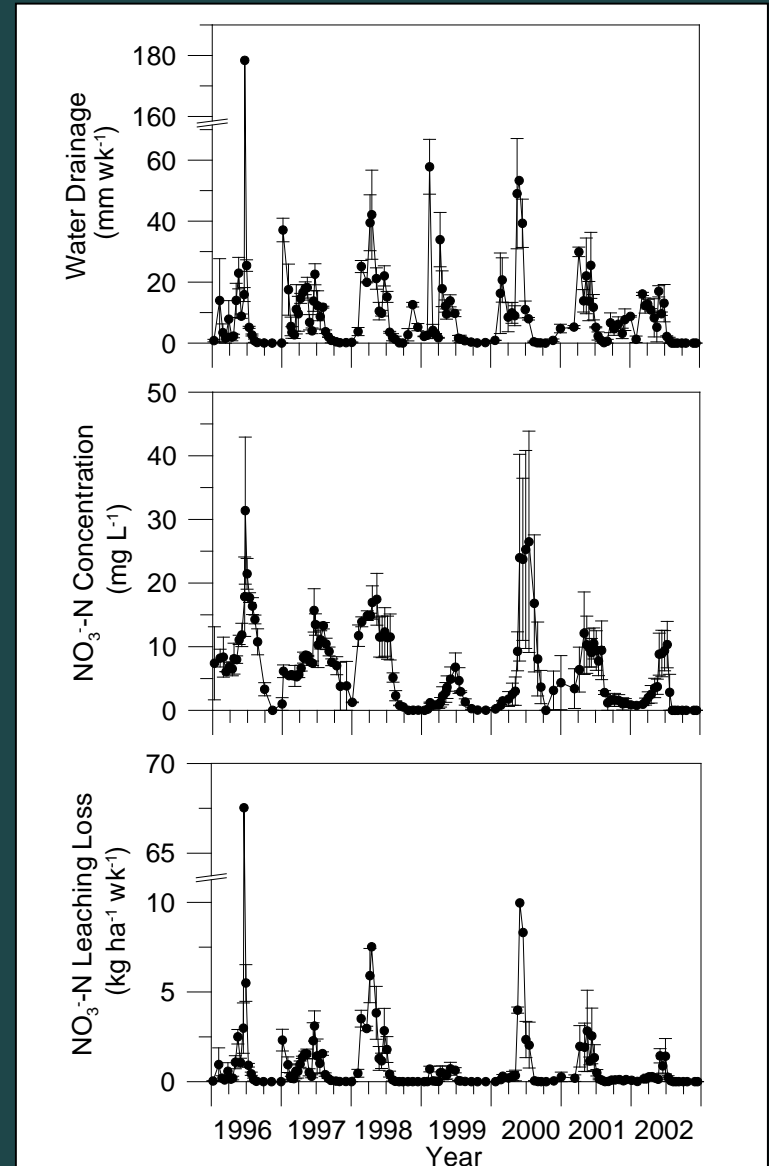


Long-term Nitrate Leaching Study

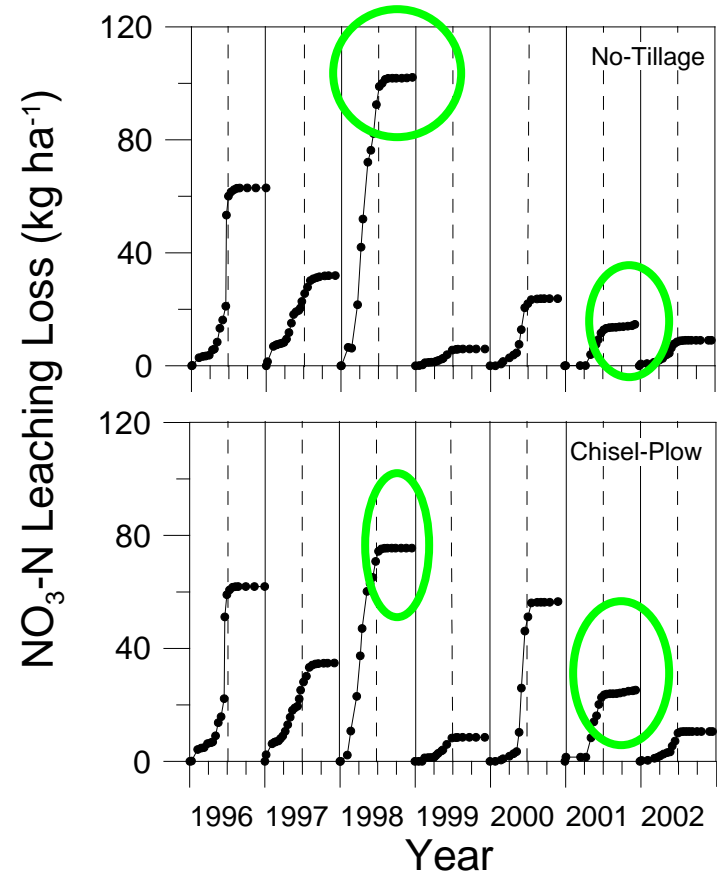
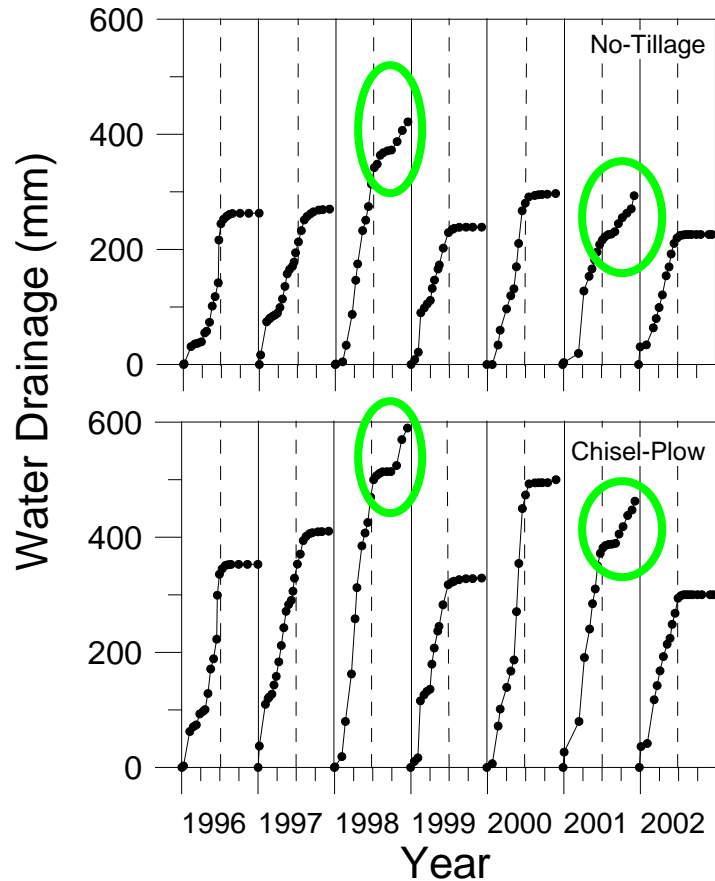
Water drainage
(mm)

$\text{NO}_3\text{-N}$ concentration
(mg L^{-1})

$\text{NO}_3\text{-N}$ leaching loss
(kg ha^{-1})



Annual Cumulative Water Drainage & Nitrate Leaching



Nitrate-impacted Municipal Wells

As of 2005 total of \$24 million

- Ø Amherst
- Ø Cambria
- Ø Chippewa Falls
- Ø Crivitz Utilities
- Ø Embarrass
- Ø Fitchburg
- Ø Fontana
- Ø Janesville Water Utility
- Ø Mattoon
- Ø Morrisonville
- Ø Oconomowoc
- Ø Orfordville
- Ø Plover
- Ø Rome
- Ø Sauk City
- Ø Strum Waterworks
- Ø Valders
- Ø Village of Arlington
- Ø Village of Clinton
- Ø Village of Dalton
- Ø Village of Footville
- Ø Village of Friesland
- Ø Waunakee
- Ø Waupaca
- Ø Whiting

What can I do to reduce my nitrate levels?

- q Possible Long-term Solution:
 - q Reduce or eliminate nitrogen inputs
- q Short term (Lewandowski et. al. 2008)
 - q Change well depth or relocate well (not guaranteed) - \$7,200
 - q Bottled water - \$190/person/year
 - q Water treatment devices - \$800 + 100/yr
 - q Reverse osmosis
 - q Distillation
 - q Anion exchange

